



*Fabrication du graphène par
exfoliation du graphite*

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Contexte - généralités

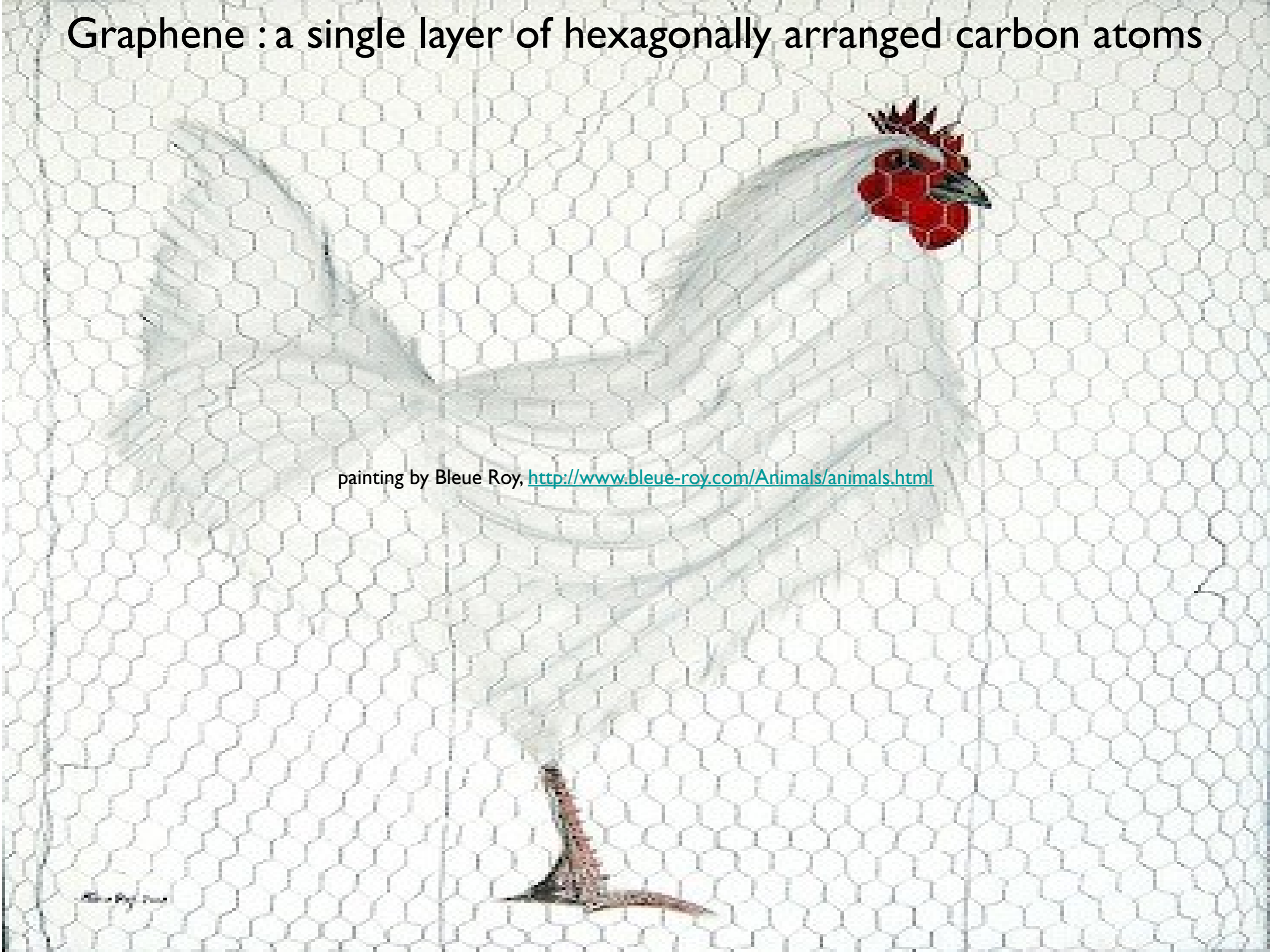
- *que désigne le terme "graphène" ?*
- *voies de synthèse du graphène*

Exfoliation du graphite (voies chimiques)

- *Oxyde de graphène (oxyde de graphite)*
- *exfoliation assistée par sonication*
- *exfoliation en voie réductrice*

Ton-scale graphene from industry ?

Graphene : a single layer of hexagonally arranged carbon atoms



painting by Bleue Roy, <http://www.bleue-roy.com/Animals/animals.html>

The ultimate membrane:

- thinnest
- strongest
- flexible yet tough
- electrically conductive

Mass less electrons: photon-like physics

- ballistic current ($l < l$ micron)
- high mobility (up to 200 000 $\text{cm}^2/\text{V.s}$)
- no electronic gap

Nobel prizes 2010

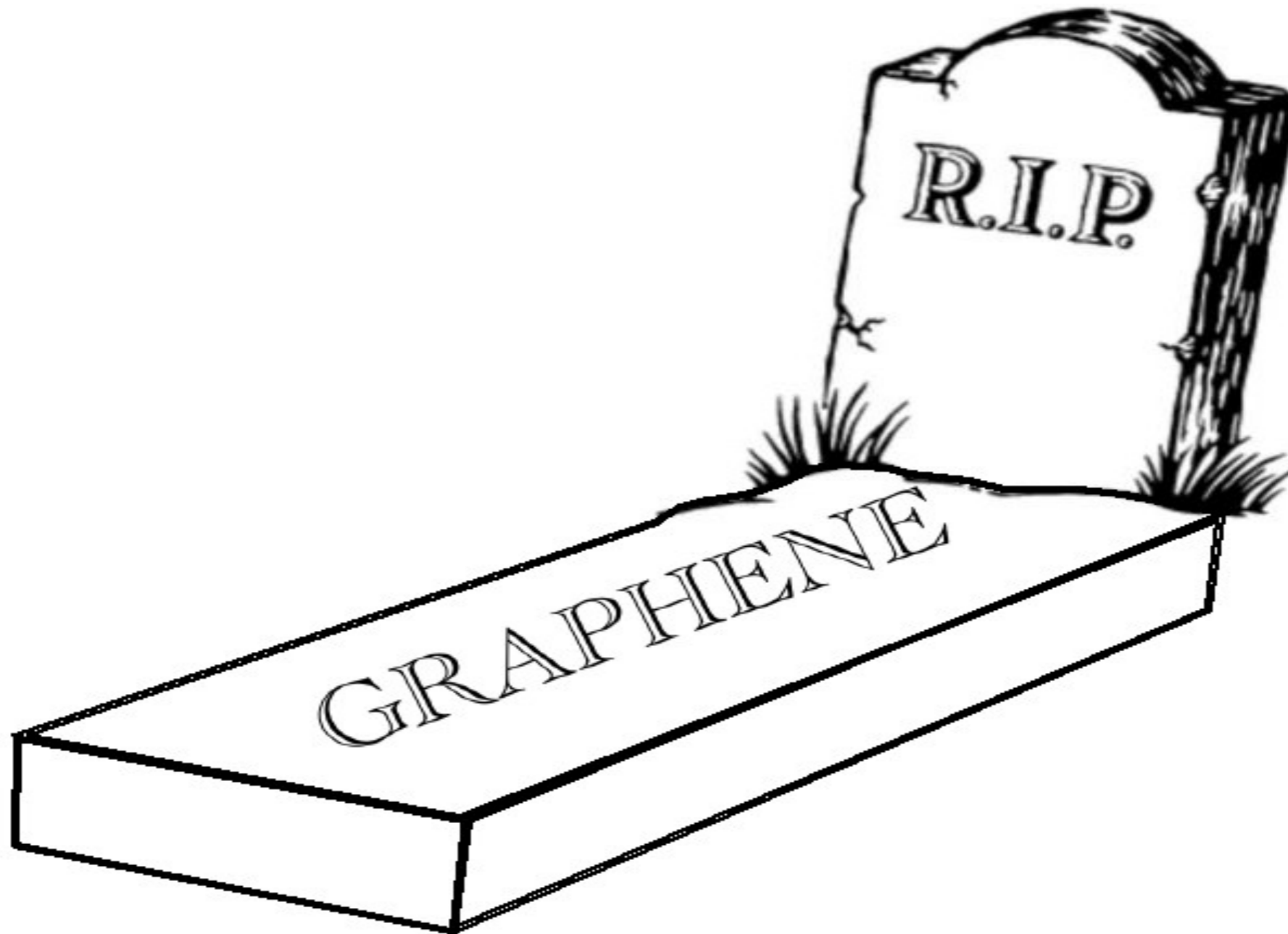
Novoselov

Geim

Diamond

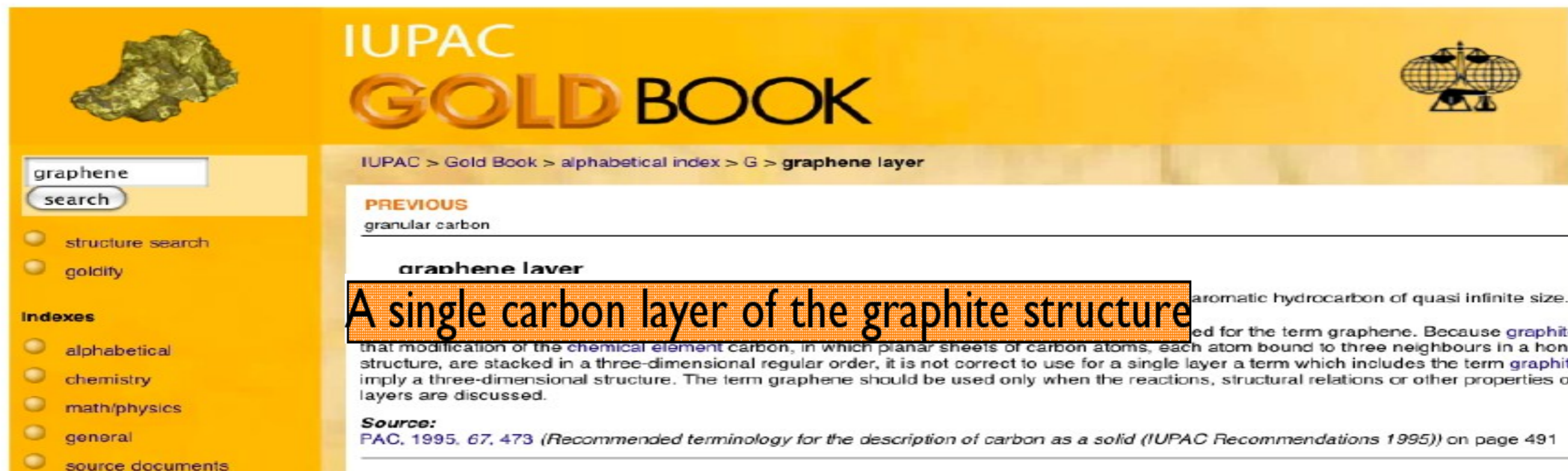


Sémantique : graphène, nanographène, nanorubans de graphène



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<http://goldbook.iupac.org/>



The screenshot shows the IUPAC Gold Book website. The main heading is "IUPAC GOLD BOOK". Below it, the breadcrumb trail reads "IUPAC > Gold Book > alphabetical index > G > graphene layer". A search bar contains the word "graphene". The main content area is titled "graphene layer" and contains the following text: "A single carbon layer of the graphite structure". Below this, there is a definition: "aromatic hydrocarbon of quasi infinite size. ... ed for the term graphene. Because graphit that modification of the chemical element carbon, in which planar sheets of carbon atoms, each atom bound to three neighbours in a honi structure, are stacked in a three-dimensional regular order, it is not correct to use for a single layer a term which includes the term graphit imply a three-dimensional structure. The term graphene should be used only when the reactions, structural relations or other properties o layers are discussed." The source is cited as "Source: PAC, 1995, 67, 473 (Recommended terminology for the description of carbon as a solid (IUPAC Recommendations 1995)) on page 491".

“Graphene is the name given to a flat monolayer of carbon atoms tightly packed into a two-dimensional (2D) honeycomb lattice”

Geim & Novoselov, Nature Materials 2007

When does 2D becomes 3D ?

... “single-, double- and few- (3 to <10) layer graphene to be distinguished as three different types of 2D crystals (‘graphenes’).”

Geim & Novoselov, Nature Materials, 2007

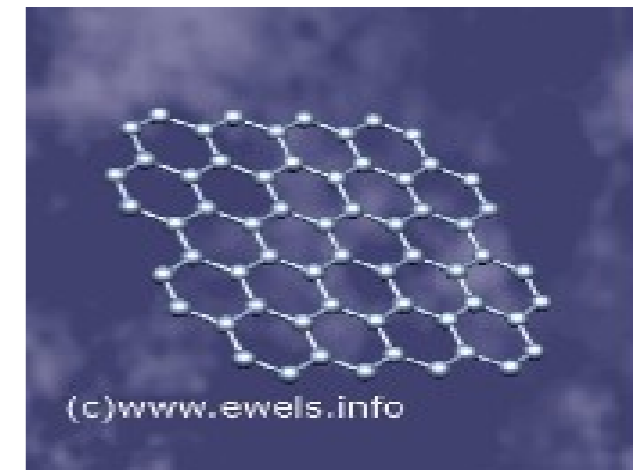
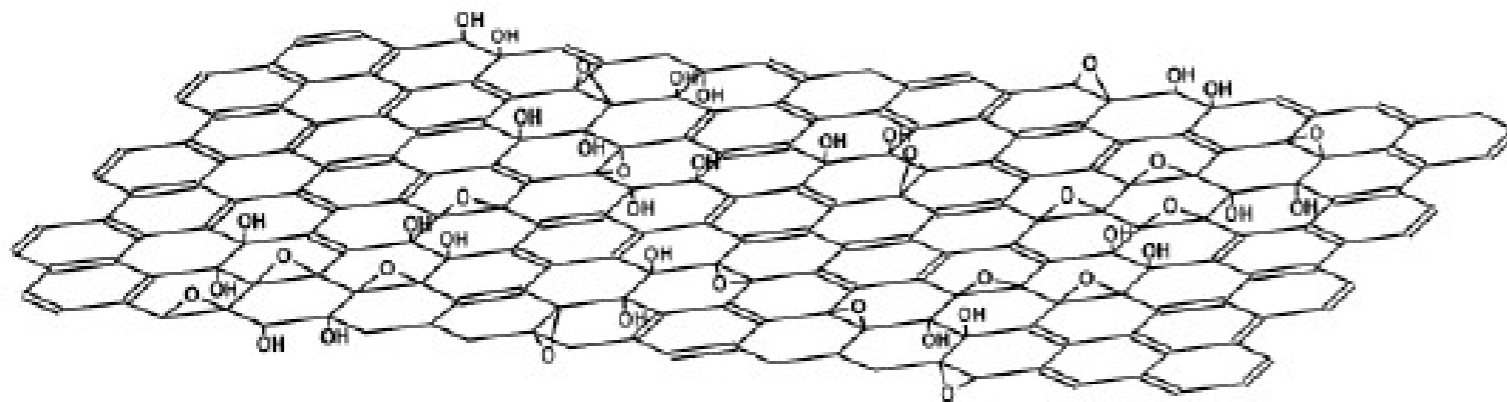
Sémantique : graphène, nanographène, nanorubans de graphène

Graphene = few (<10) graphitic layers

Few Graphene Layers

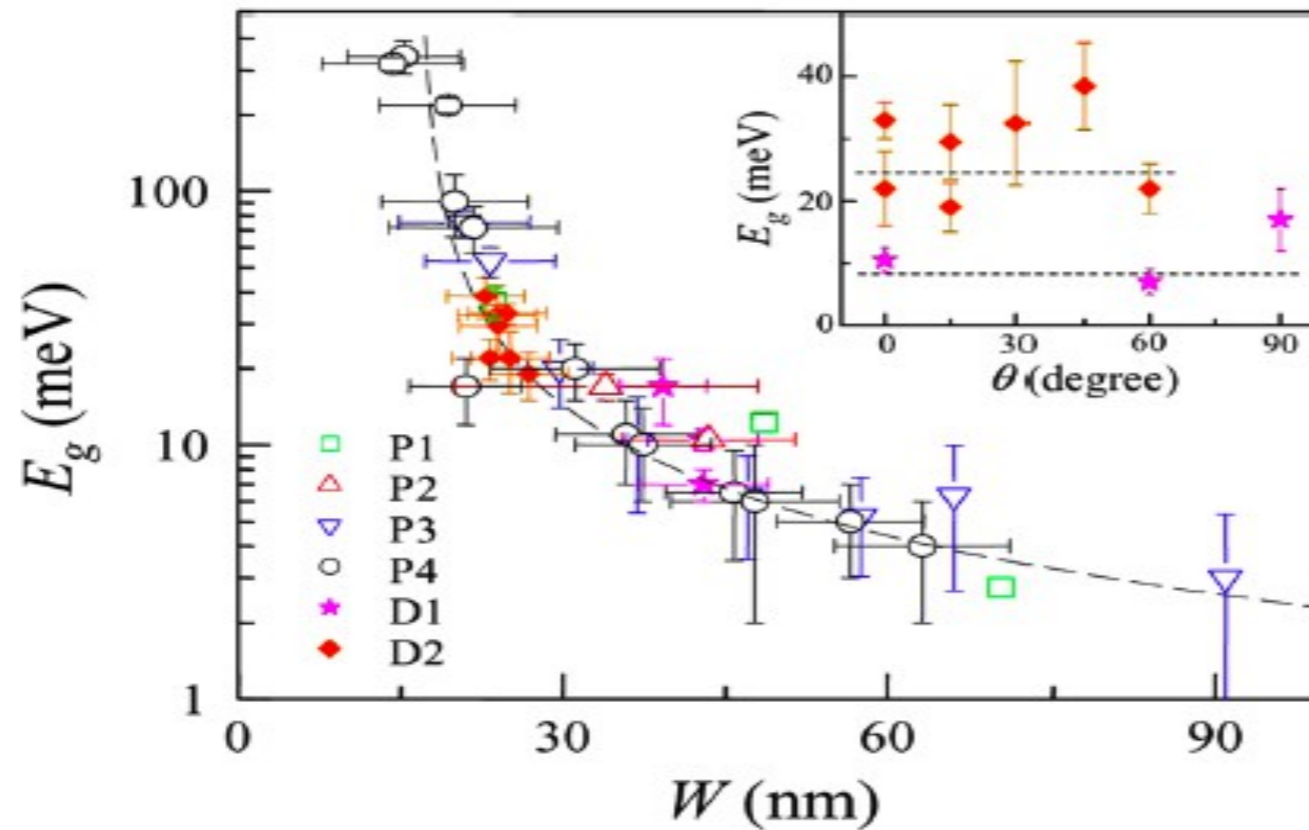
Sémantique : graphène, nanographène, nanorubans de graphène

Graphene = Oxyde de graphène réduit

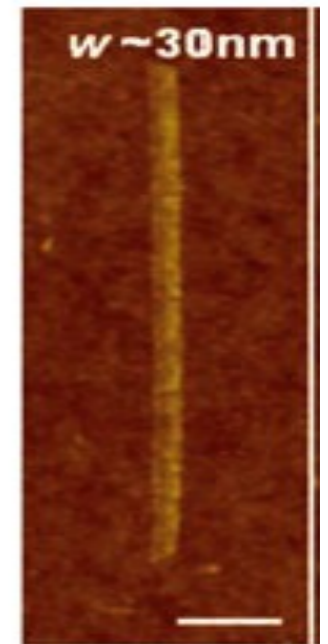


Sémantique : graphène, nanographène, nanorubans de graphène

Graphene nanoribbons: graphene is already “nano” in thickness :
“nano” should apply to width



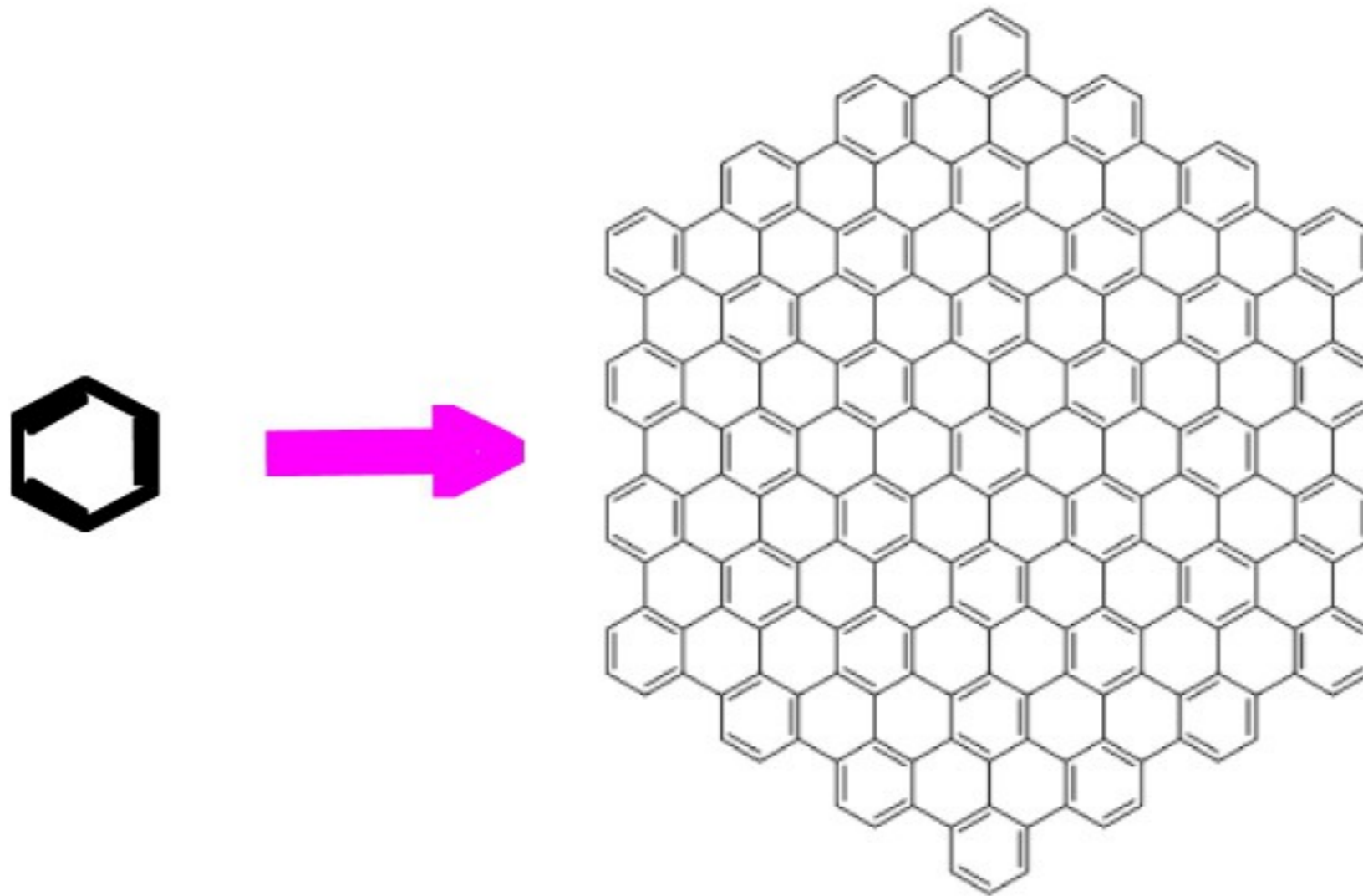
Kim et al., PRL, 2007



H. Dai et al., Science 2008

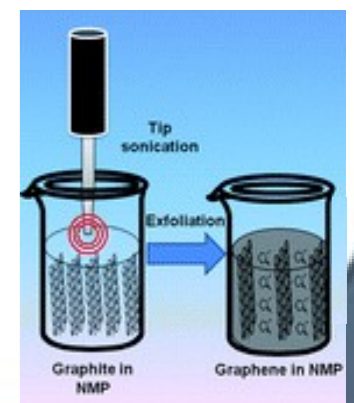
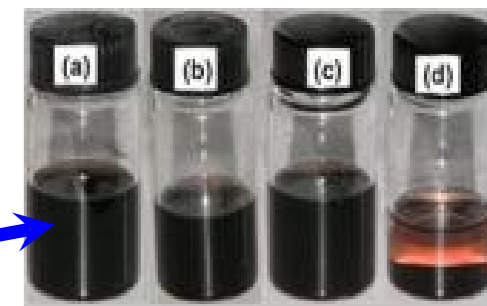
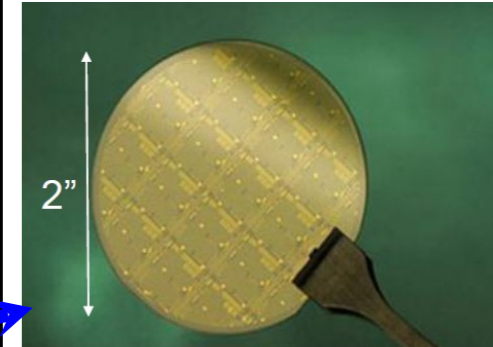
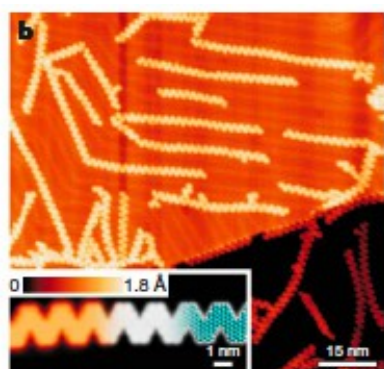
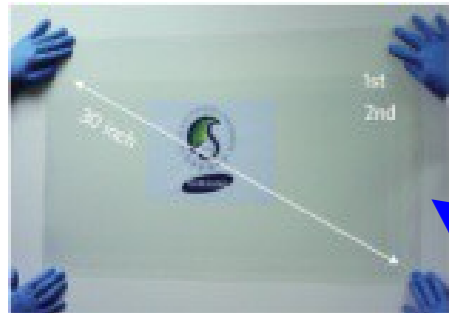
Sémantique : graphène, nanographène, nanorubans de graphène

Where PolyAromatic Hydrocarbons (PAHs) and Graphene meet...



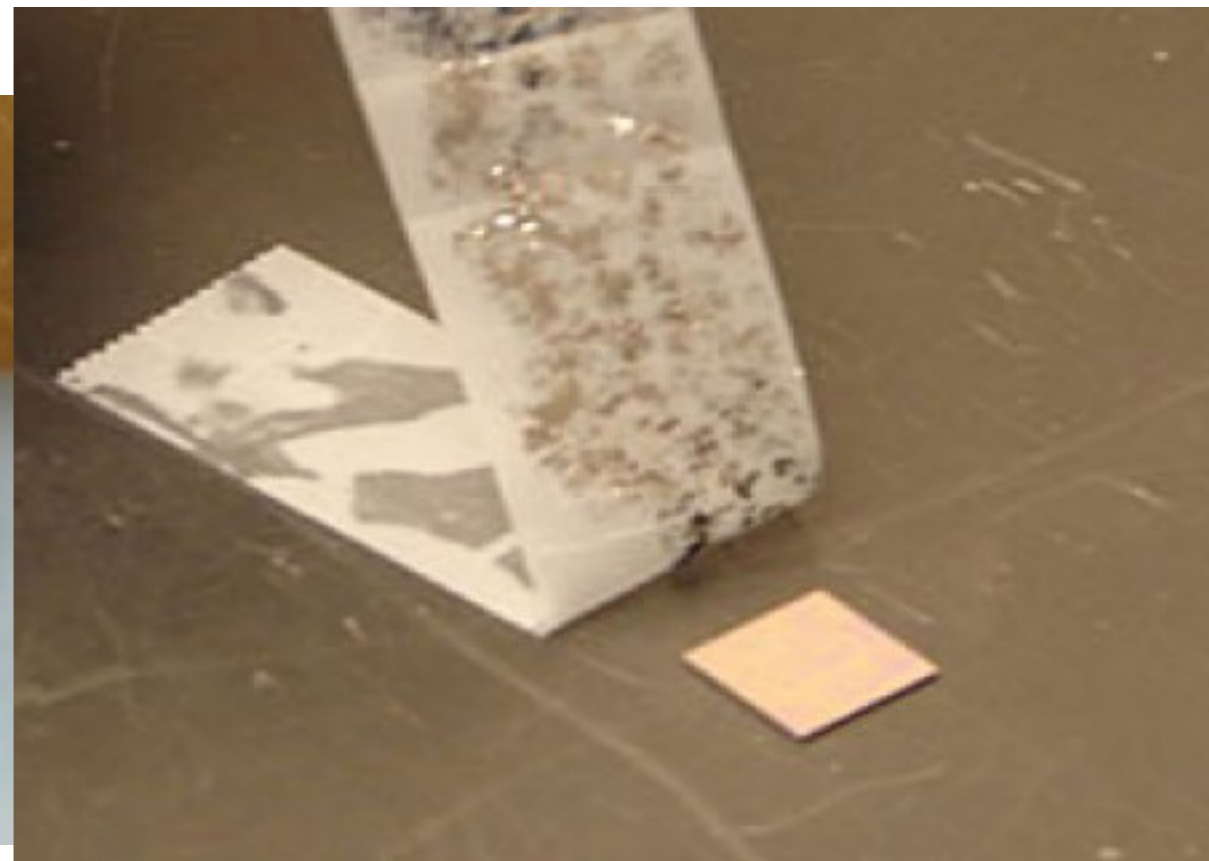
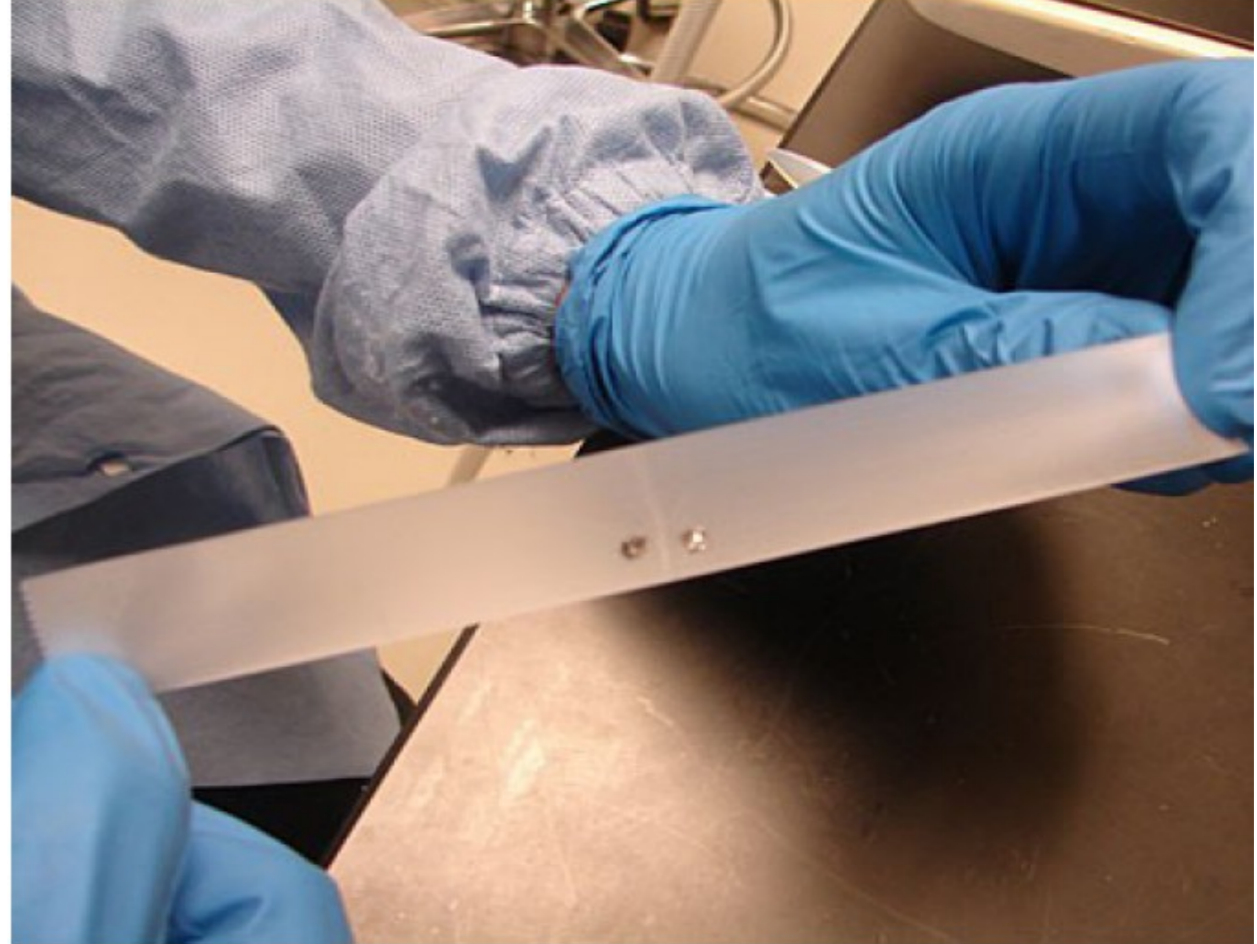
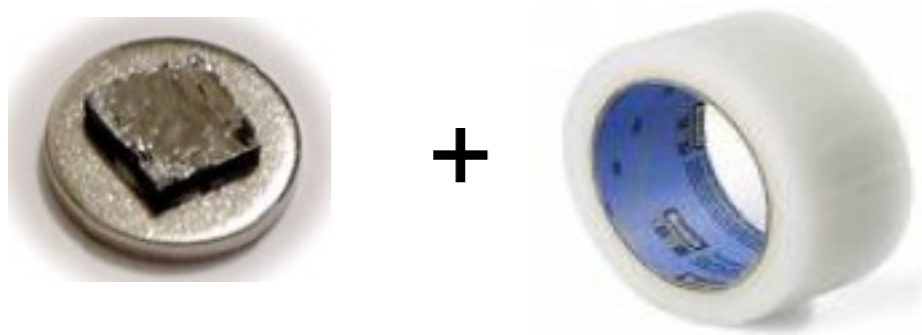
Graphene production methods

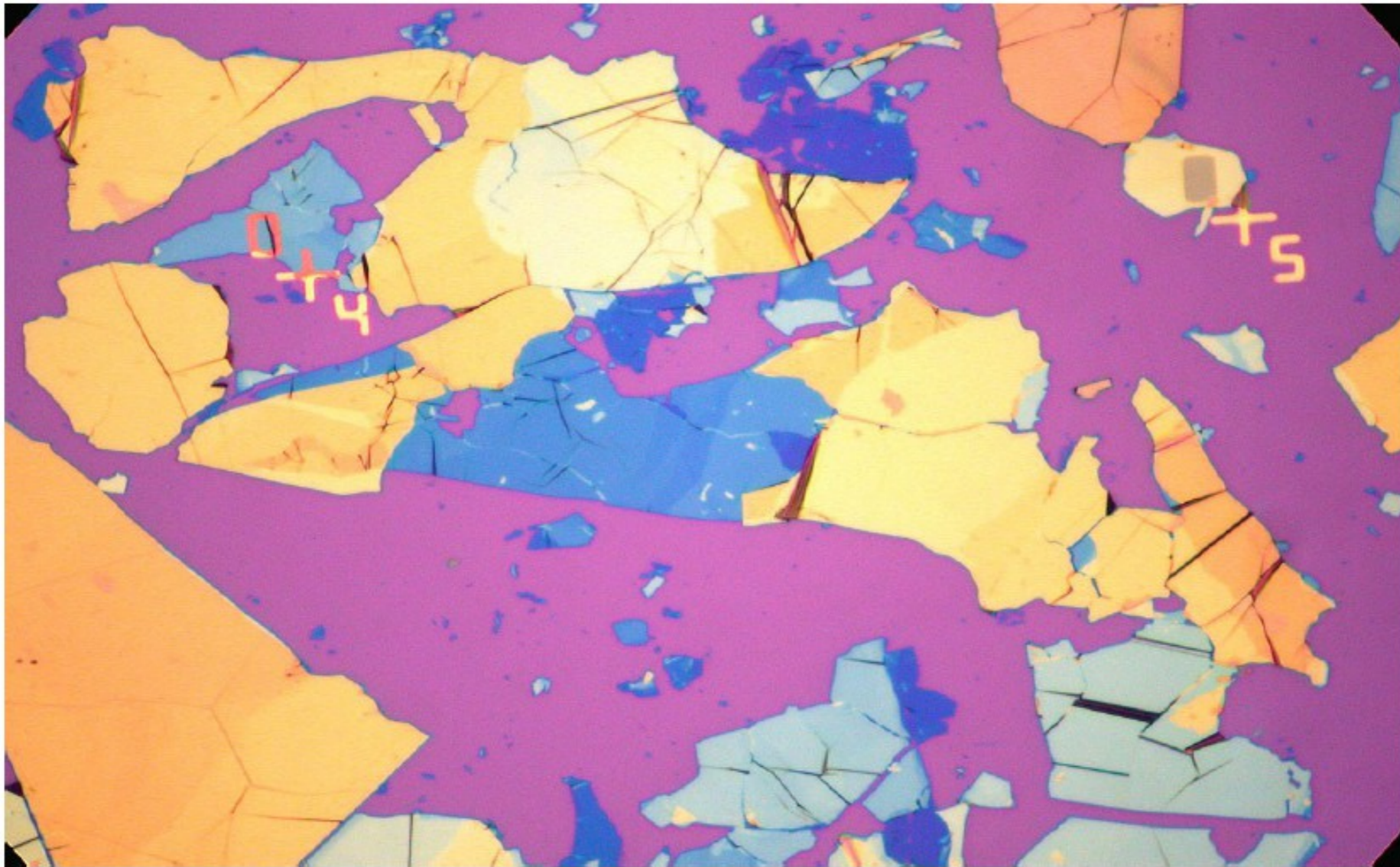
	☺	☹
Mechanical exfoliation	• Quality	• Low yield • High costs
Epitaxial growth on SiC	• Quality, • Large area	• Substrat • high cost
Chemical Vapor Deposition	• Large area	• High costs • Substrat
CNT unzipping	• ribbons	• quality, cost
Total synthesis	• monodisperse size	• quantities, cost
Graphite Oxide Reduction	• Large scale	• Quality reduced
Dispersion of Graphite	• Large scale	• partial exfoliation



→ Solution of Graphite intercalation Compounds (GICs) in organic solvents

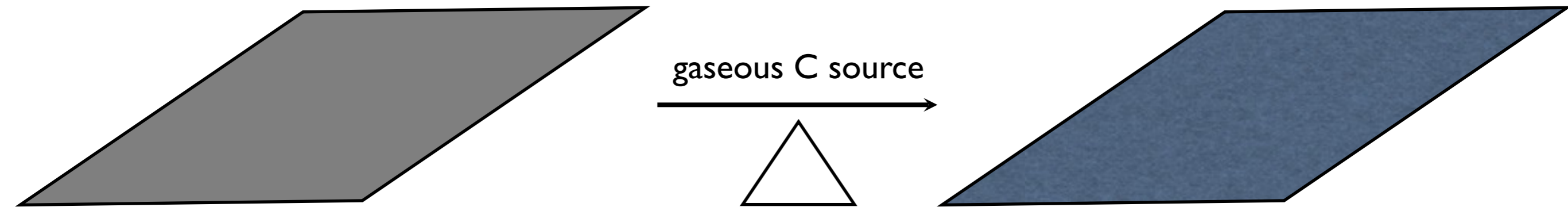
slide inspired from J.N. Fuchs





slide inspired from J.N. Fuchs

graphene by CVD

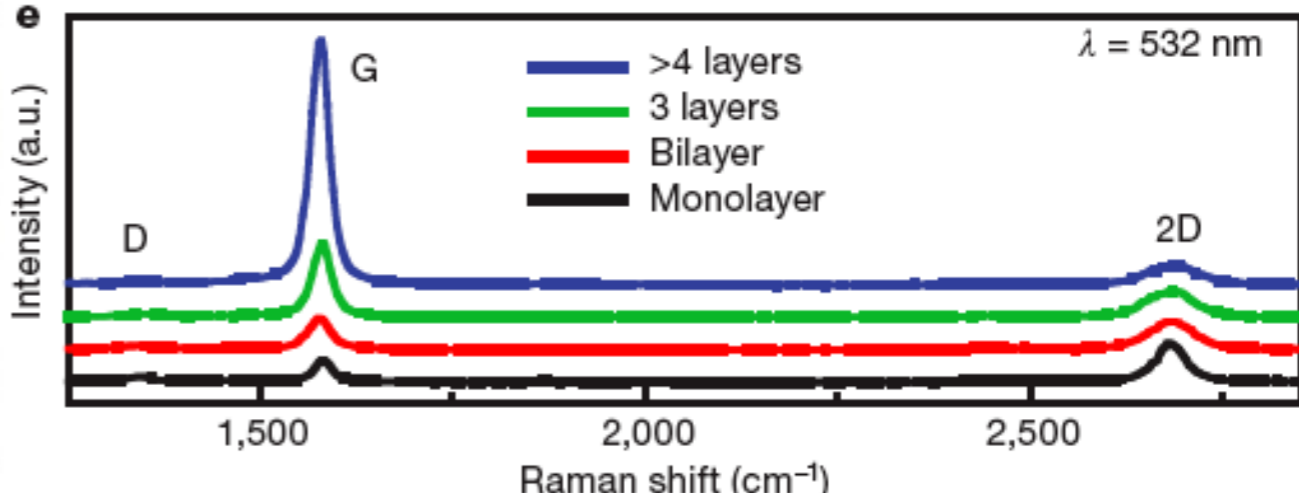
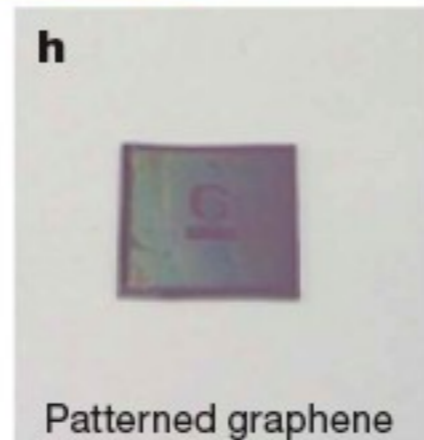
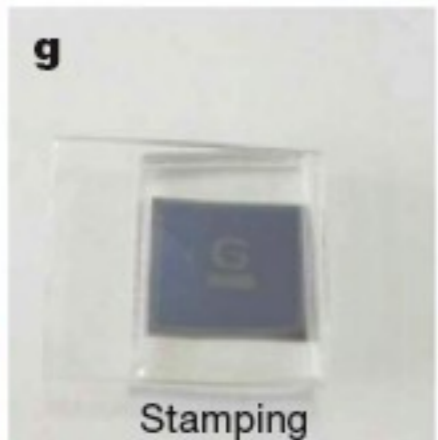
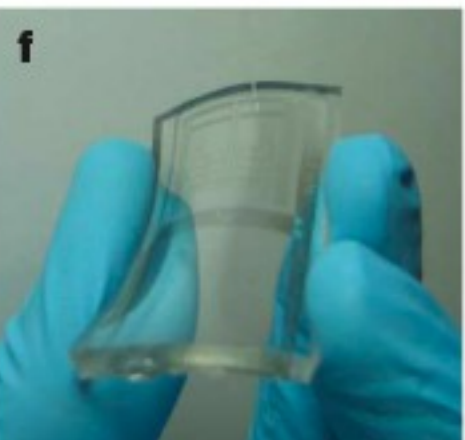
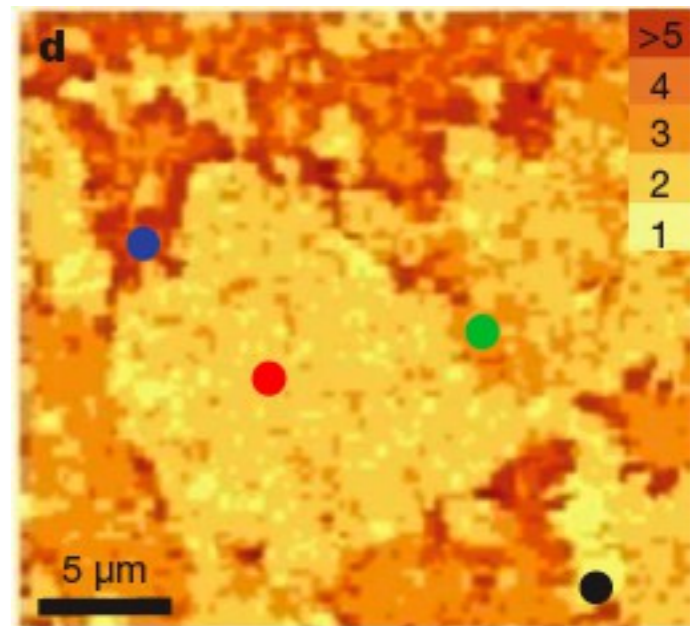
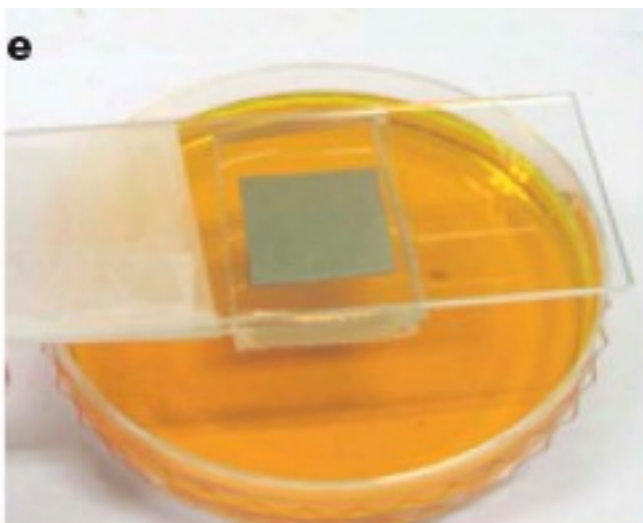
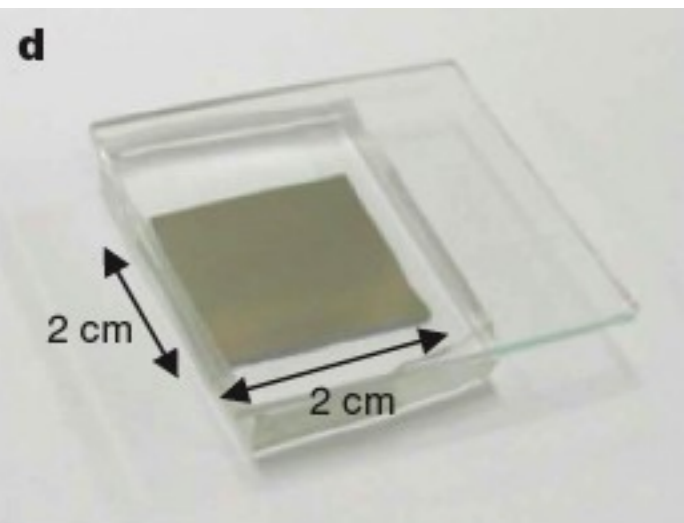
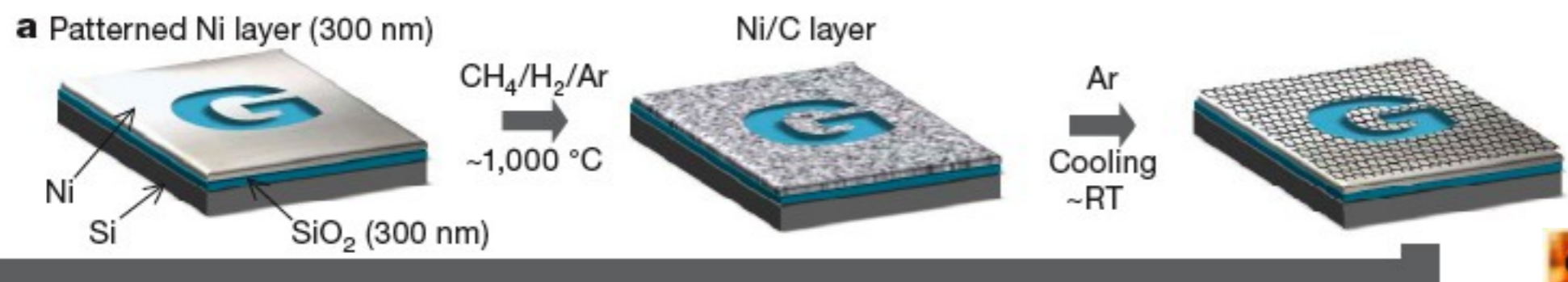


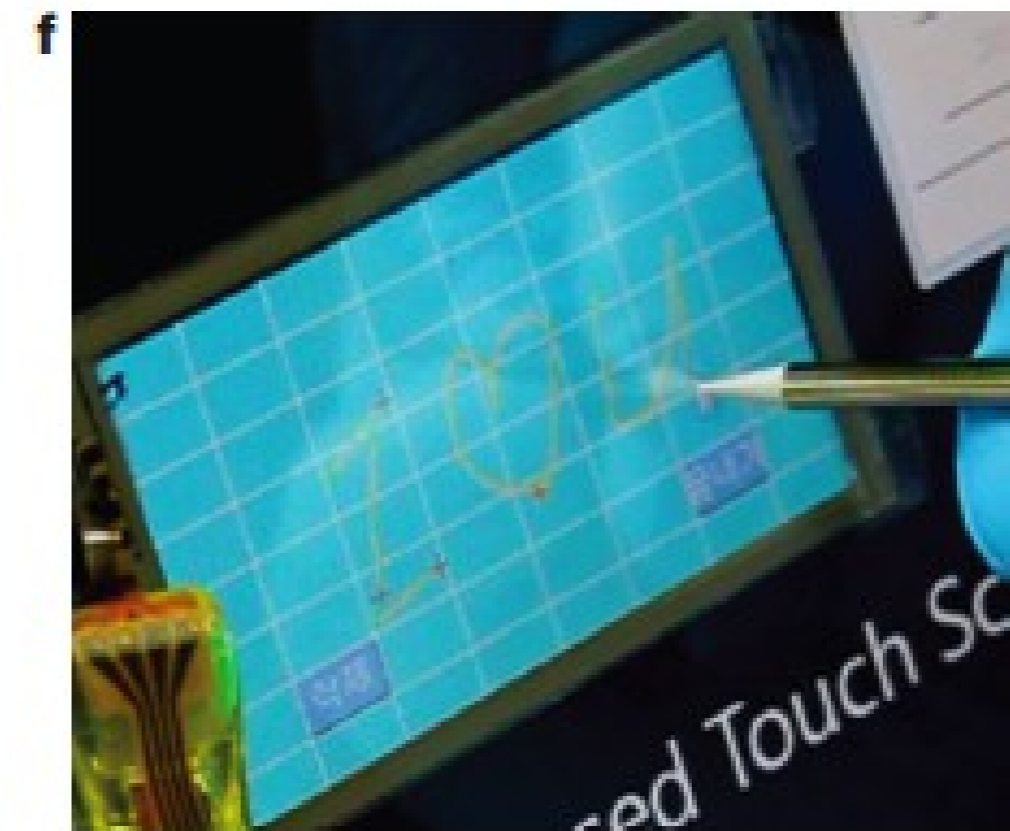
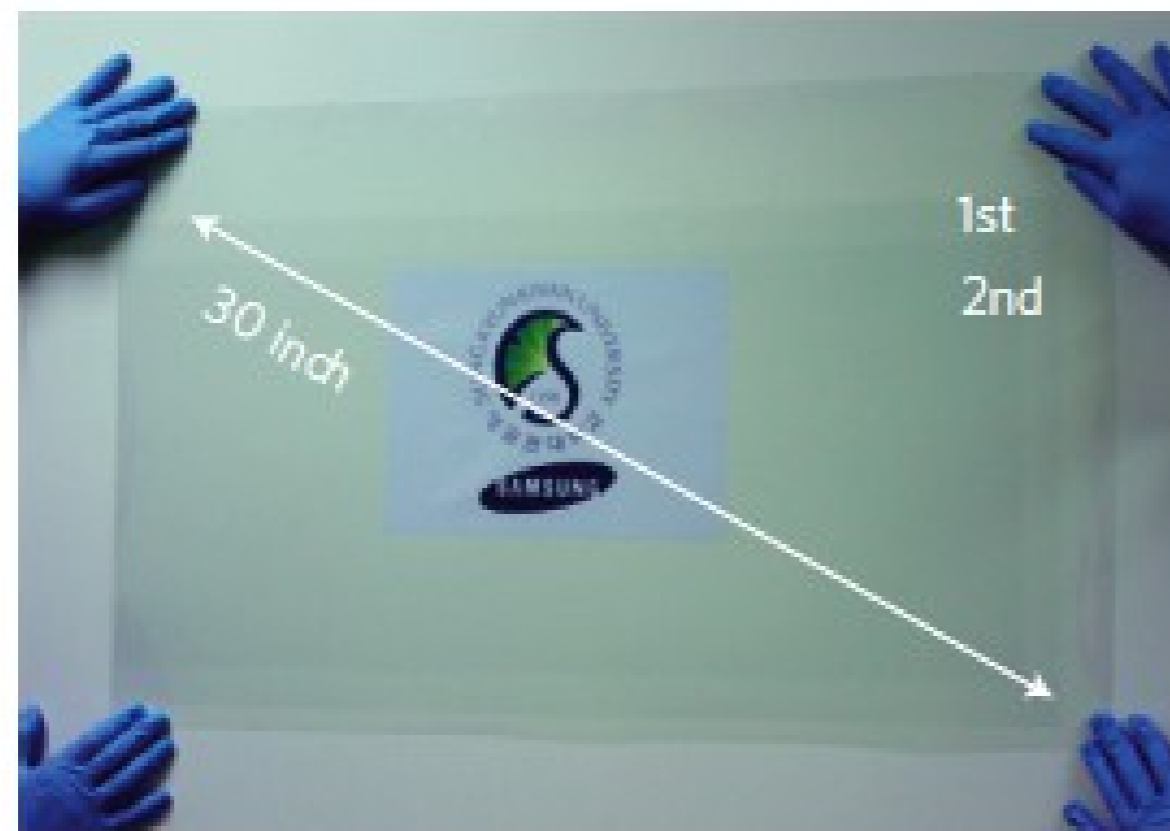
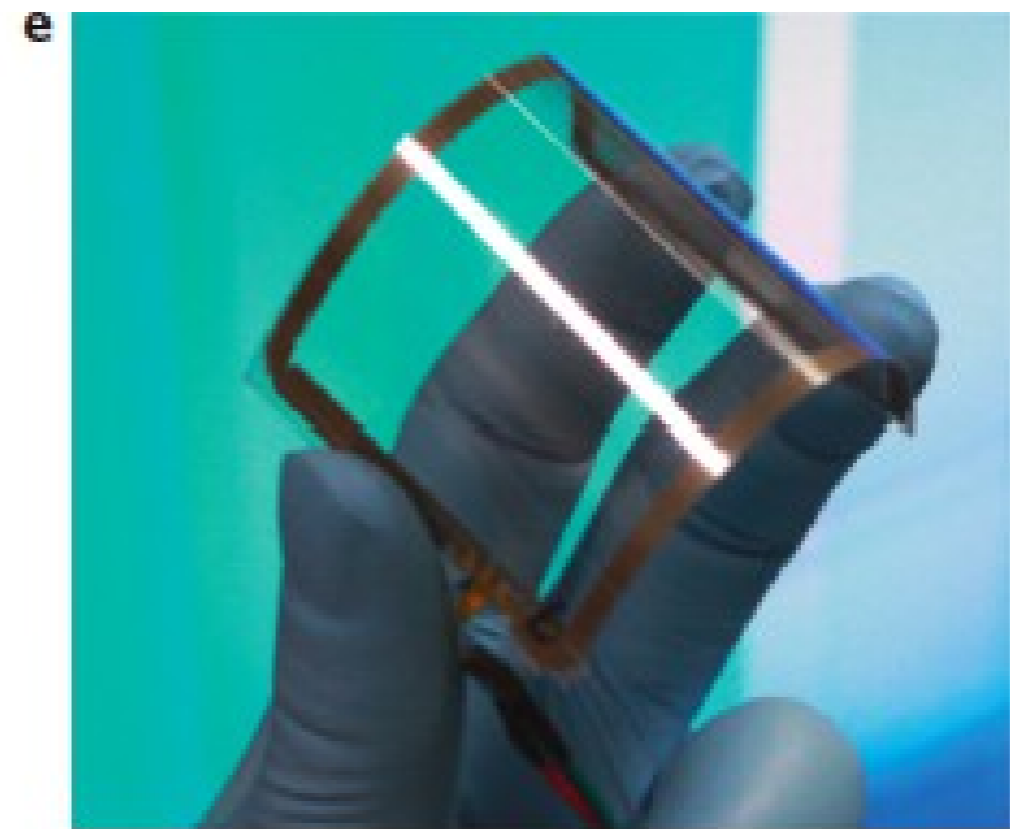
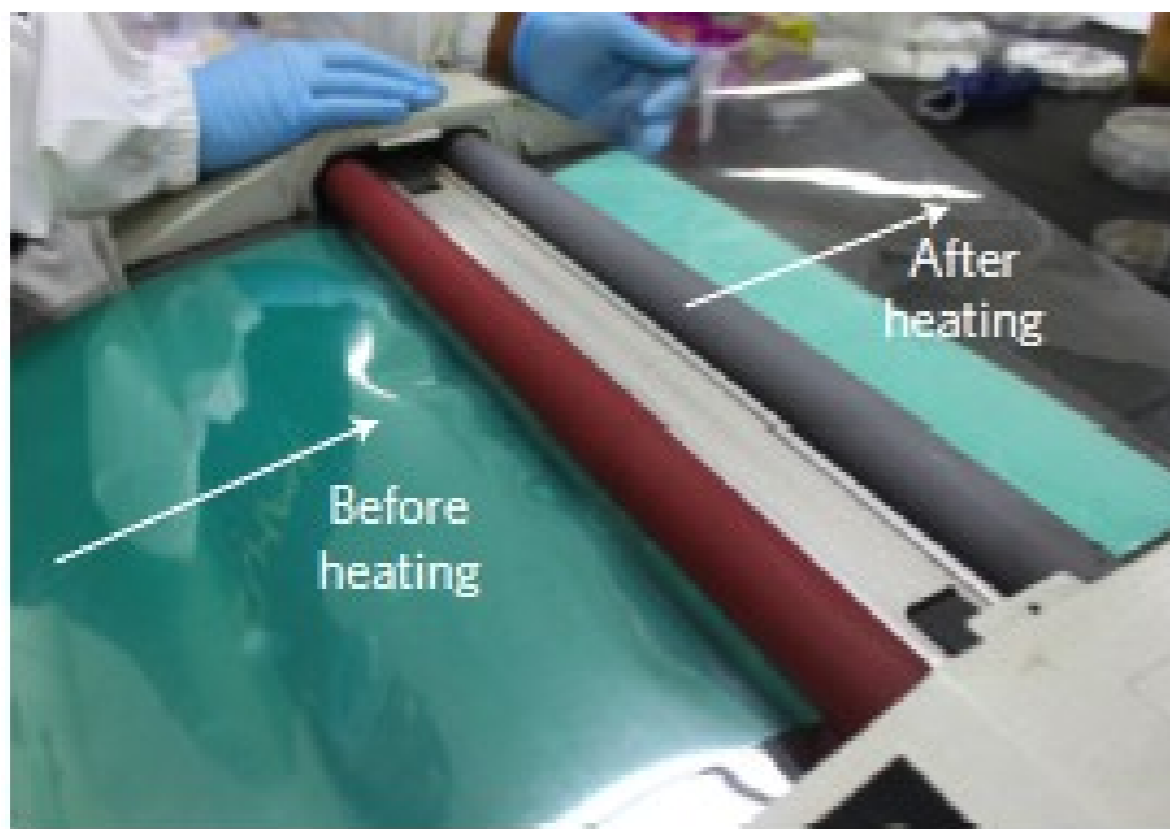
etching of the substrate
or stamping

Large-scale pattern growth of graphene films for stretchable transparent electrodes

Keun Soo Kim^{1,3,4}, Yue Zhao⁷, Houk Jang², Sang Yoon Lee⁵, Jong Min Kim⁵, Kwang S. Kim⁶, Jong-Hyun Ahn^{2,3}, Philip Kim^{3,7}, Jae-Young Choi⁵ & Byung Hee Hong^{1,3,4}

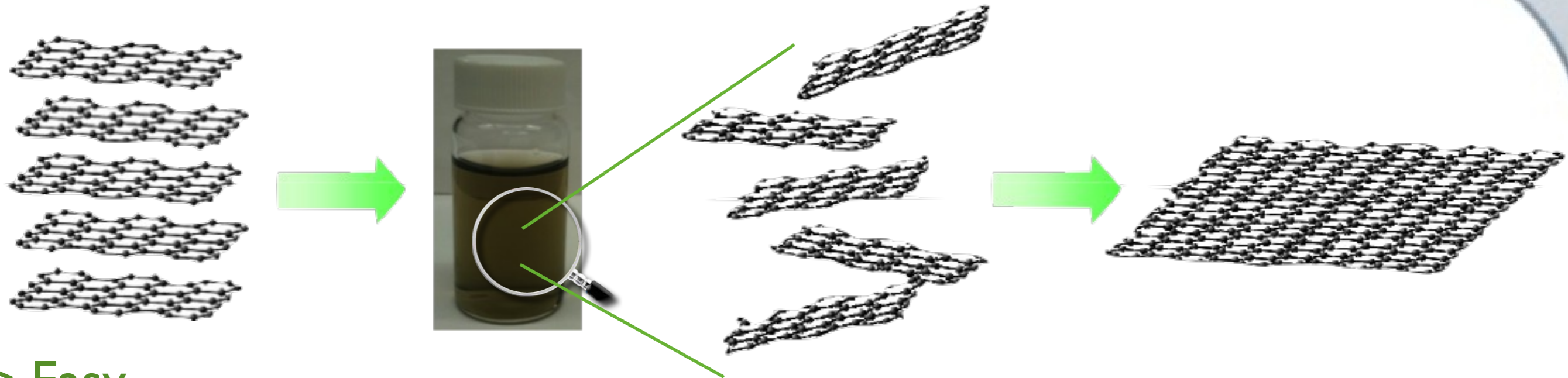
Nature, 14 Janvier 2009



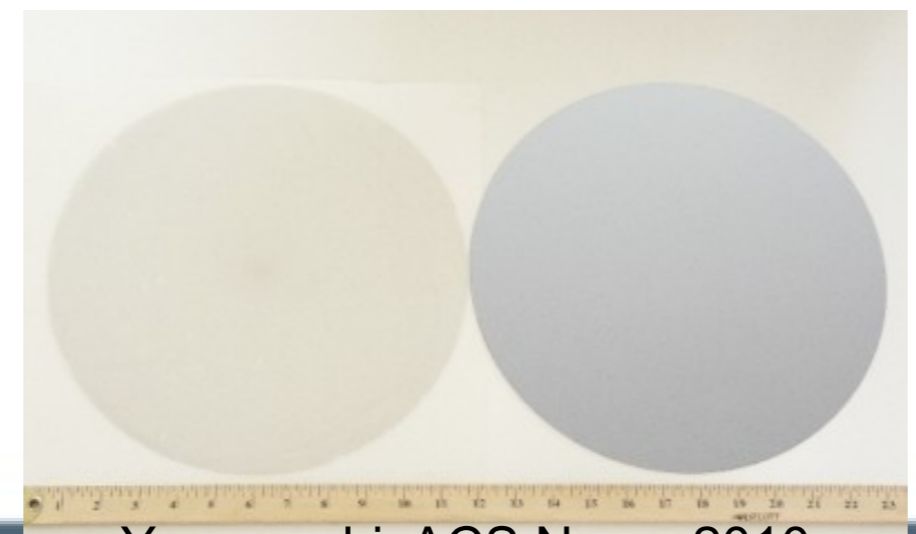
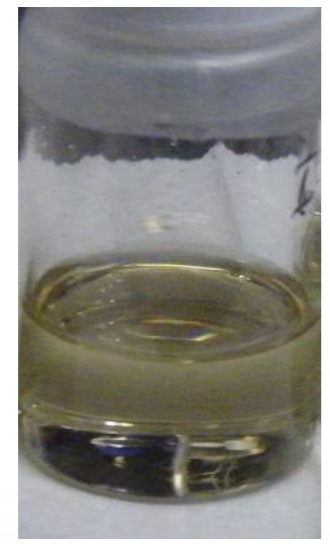
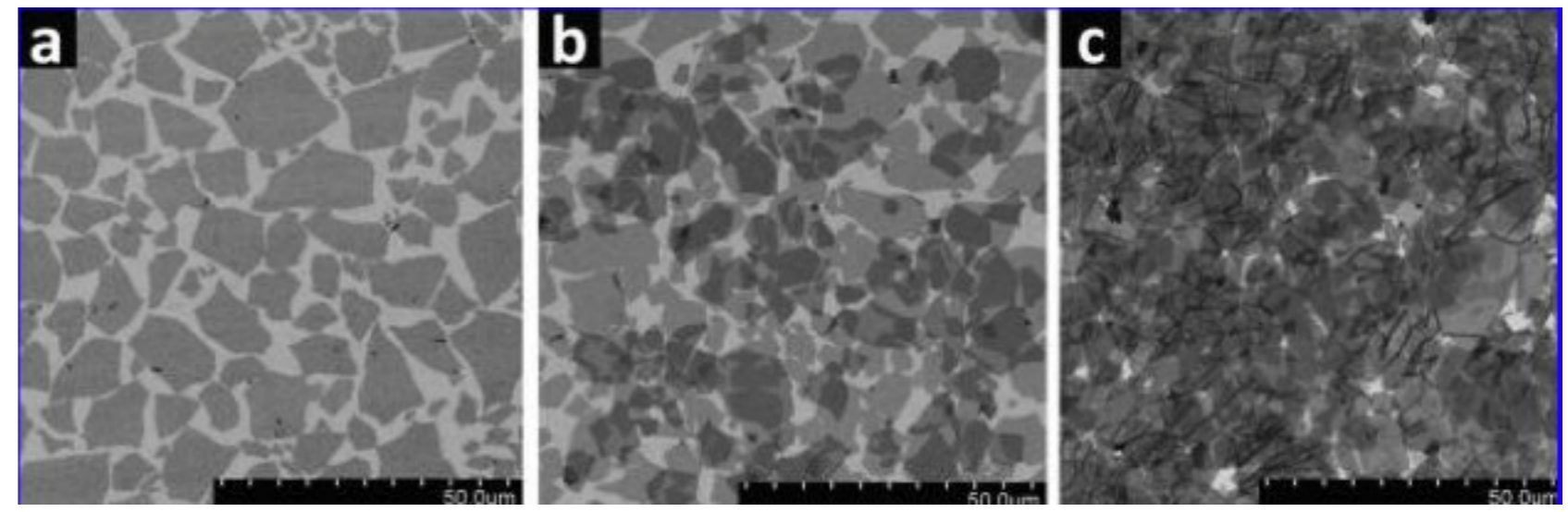


Byung Hee Hong et al., Nature Nanotechnology, 2010

Why solution process?



- > Easy
- > Cheap
- > Large quantity
- > Scalable
- > Chemical modifications



Cote, JACS, 2008

Yamaguchi, ACS Nano, 2010

Which graphite ?

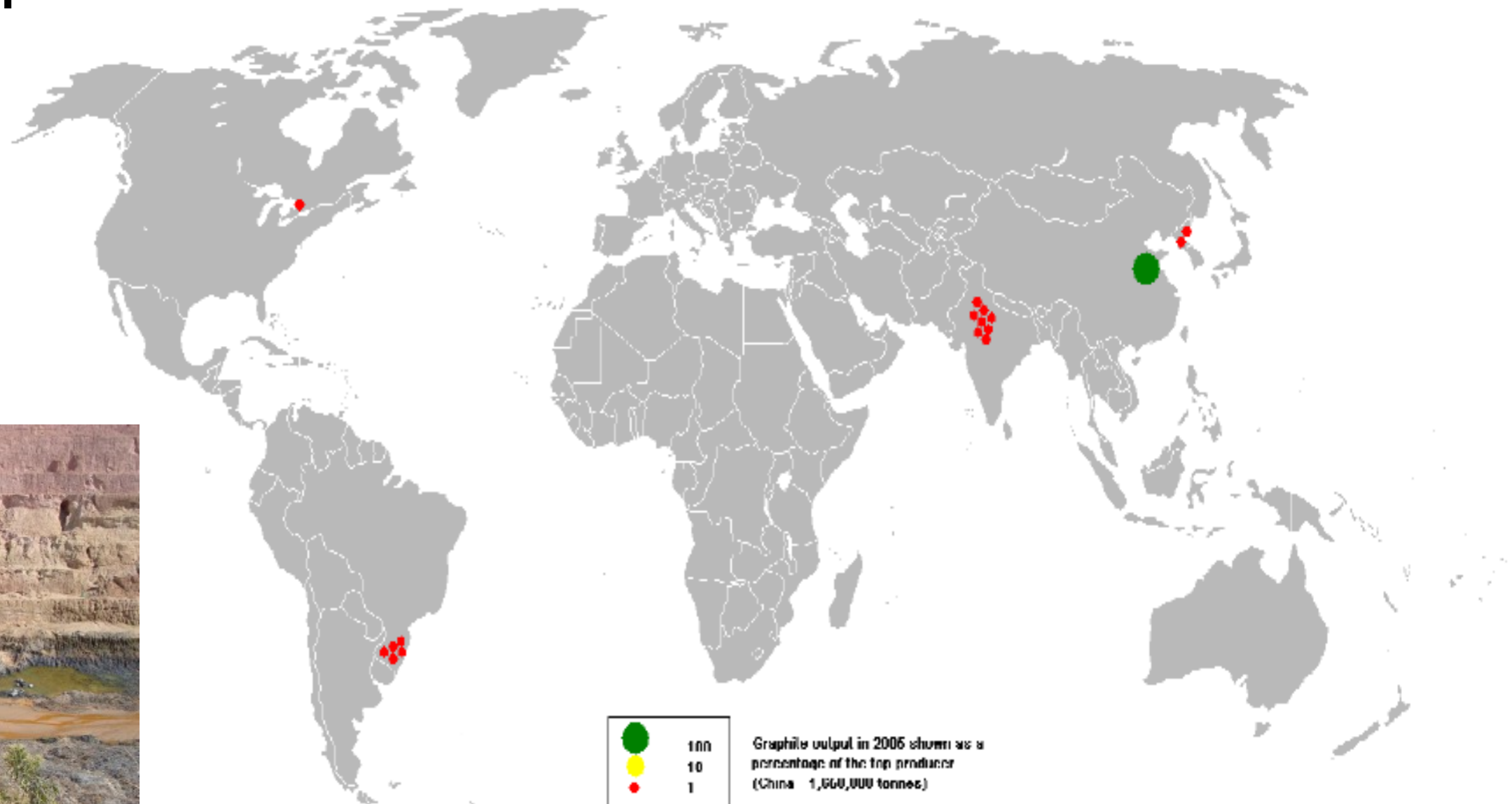
Natural graphite
(extracted from
mines)

Kish graphite (recycled of the steel industry)

Highly Oriented Pyrolytic Graphite (HOPG)

Natural graphite

Expanded graphite



Natural graphite



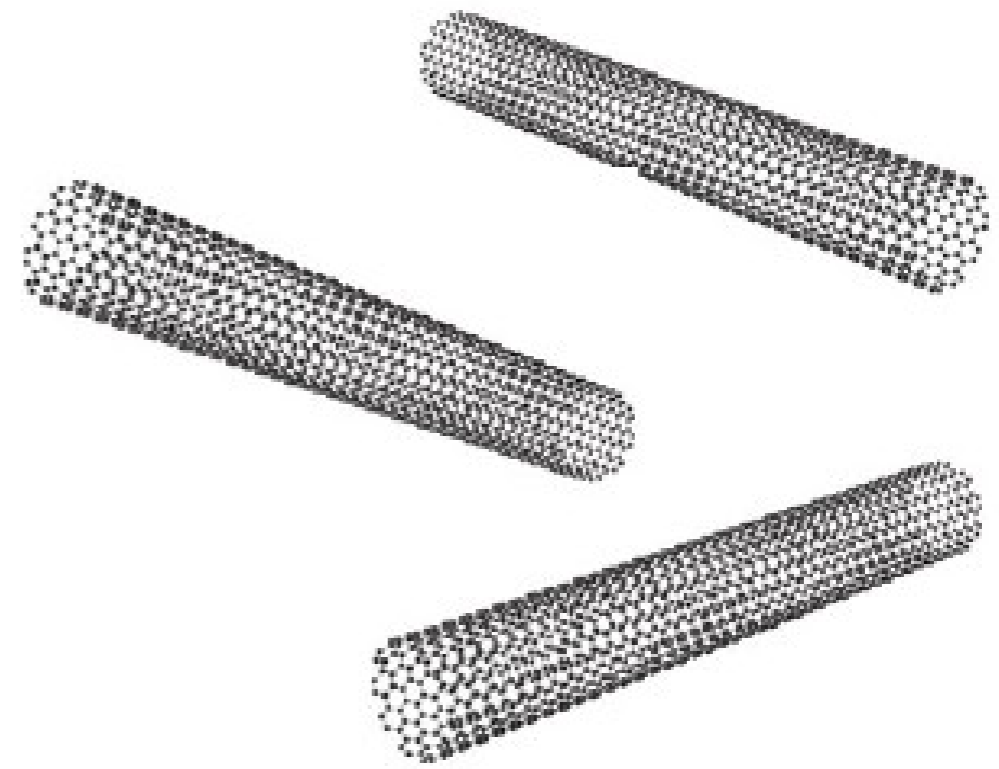
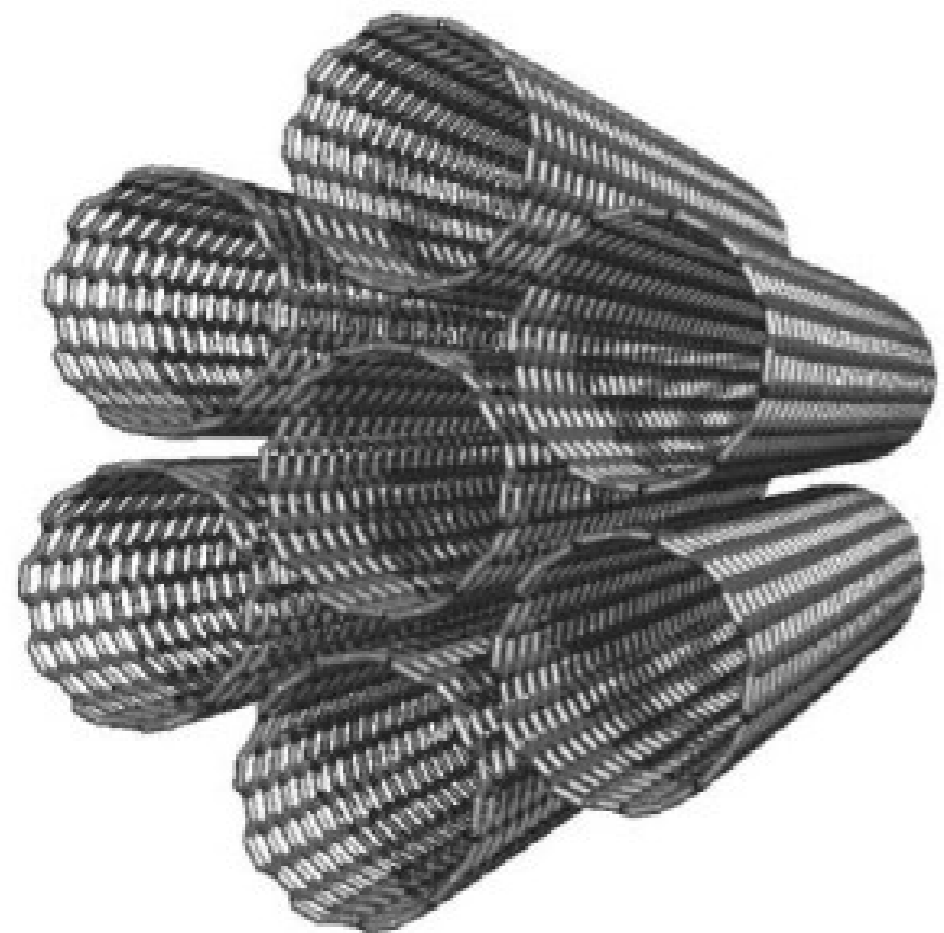
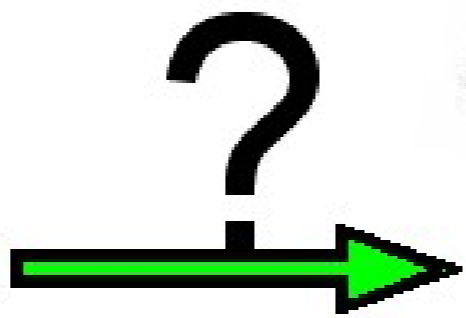
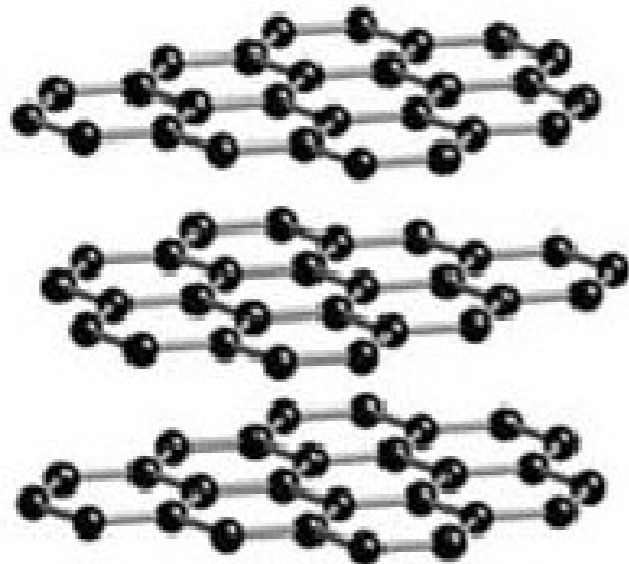
Graphite mine in Minas Gerais, Nacional de Grafite, Brazil, (photo by A. Pénicaud)



Photo N. Rosas, nacional de Grafite



Bags of graphite, Nacional de Grafite, Brazil, (photo by A. Pénicaud)



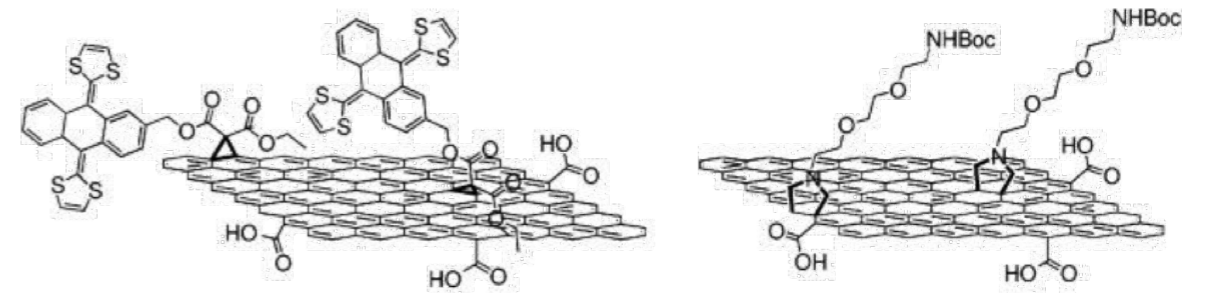
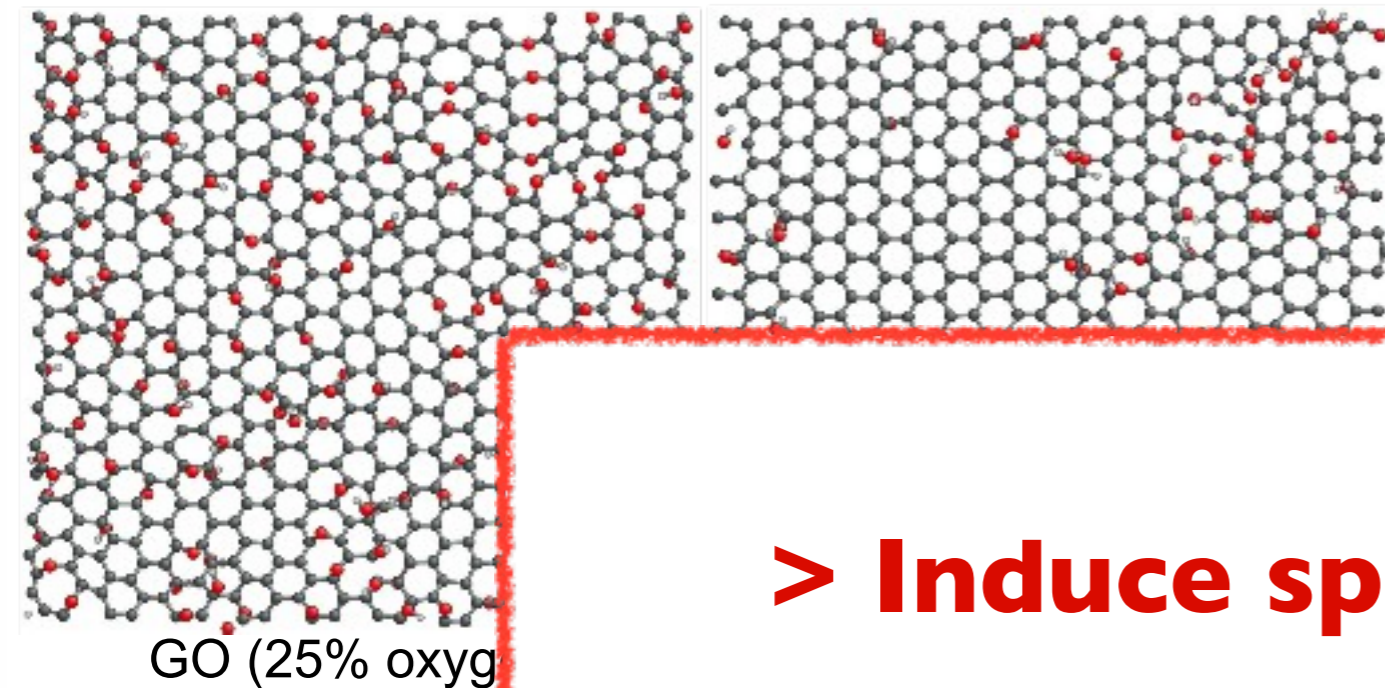
Solution process

Chemical treatment:

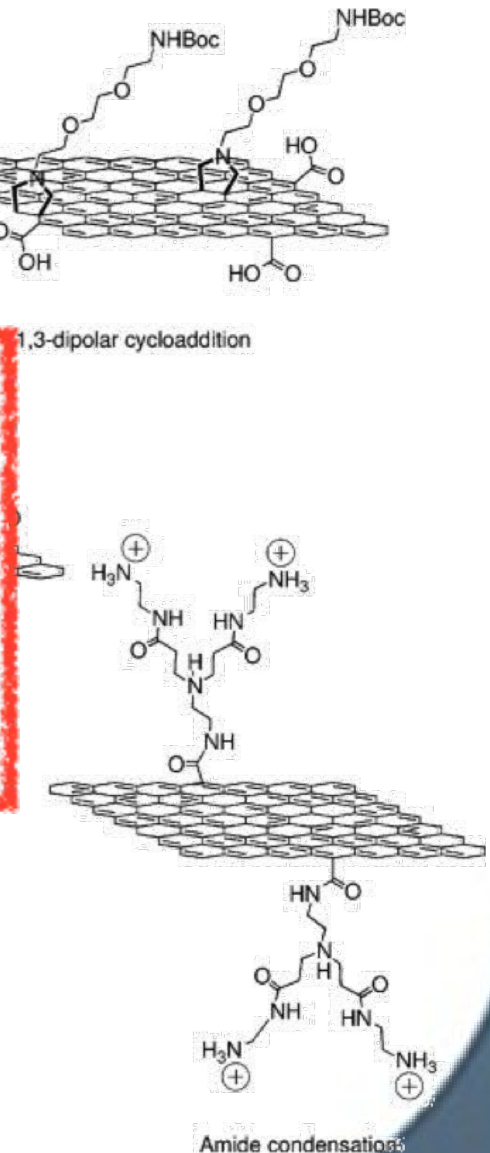
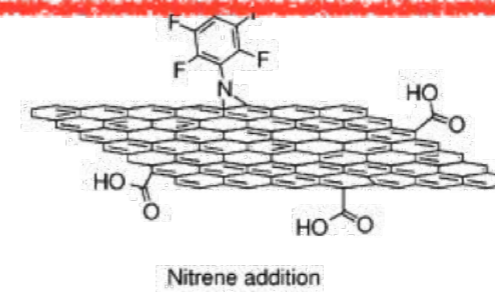
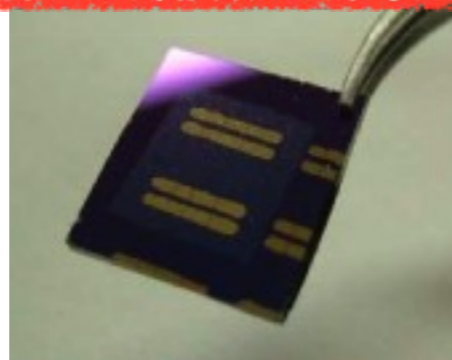
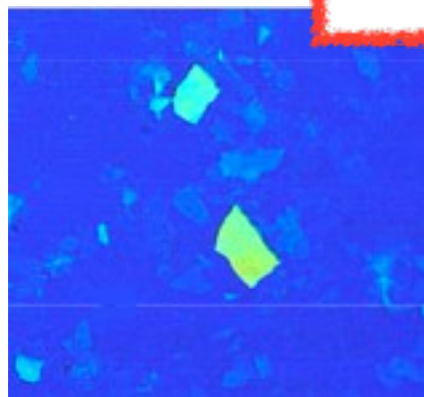
Graphene oxide
(known since 1859!)

Functionalization
of dispersed graphene

(based on reactions developed for CNTs)



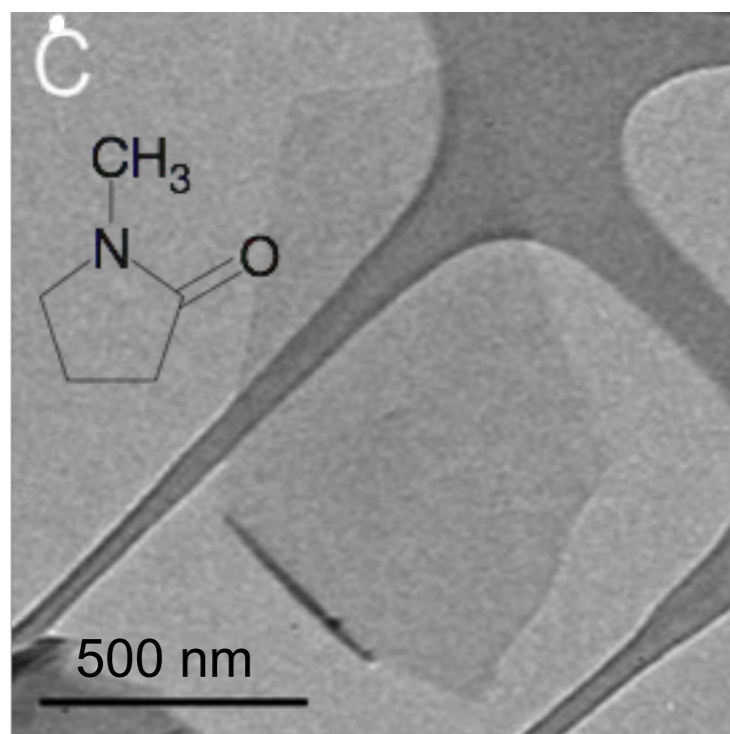
> Induce sp^3 defects...



SOLUTION PROCESS

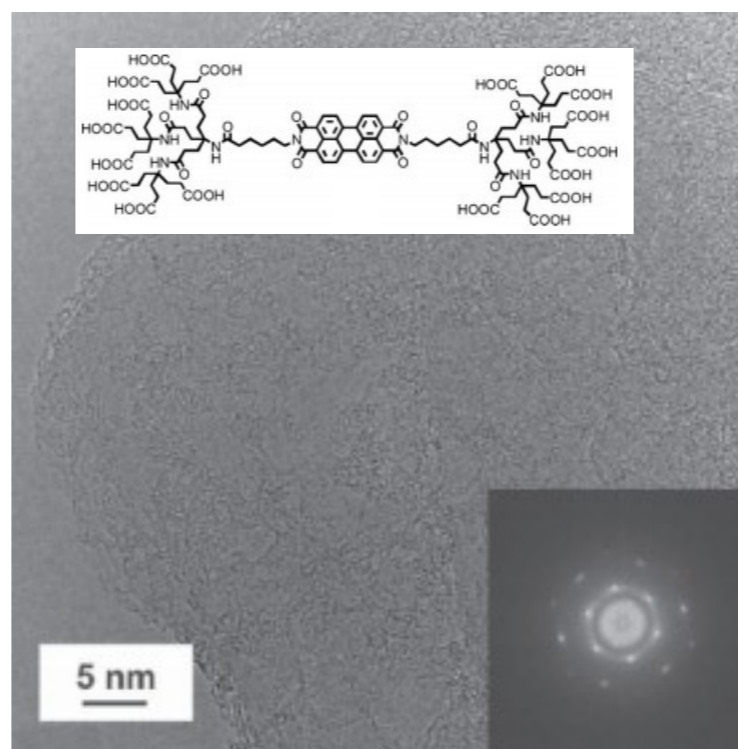
Dispersion:

Organic solvent

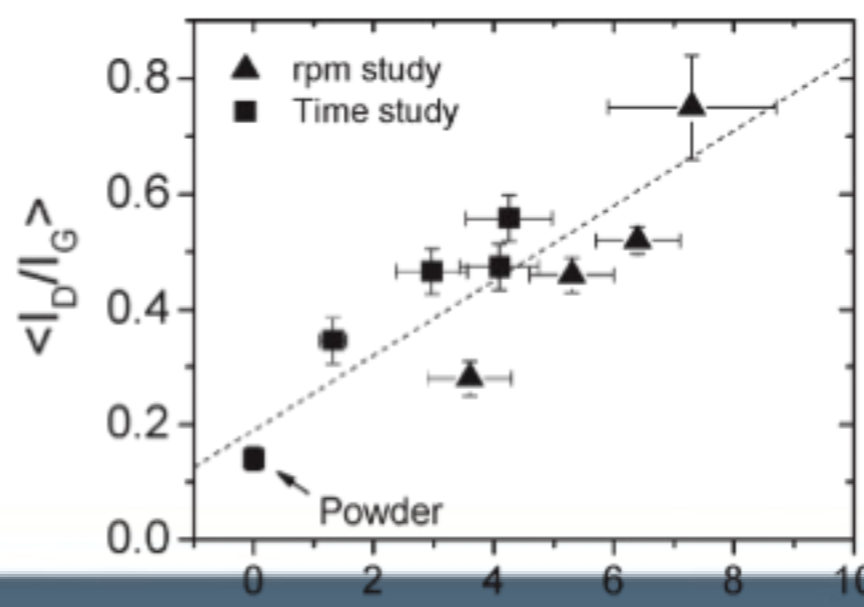
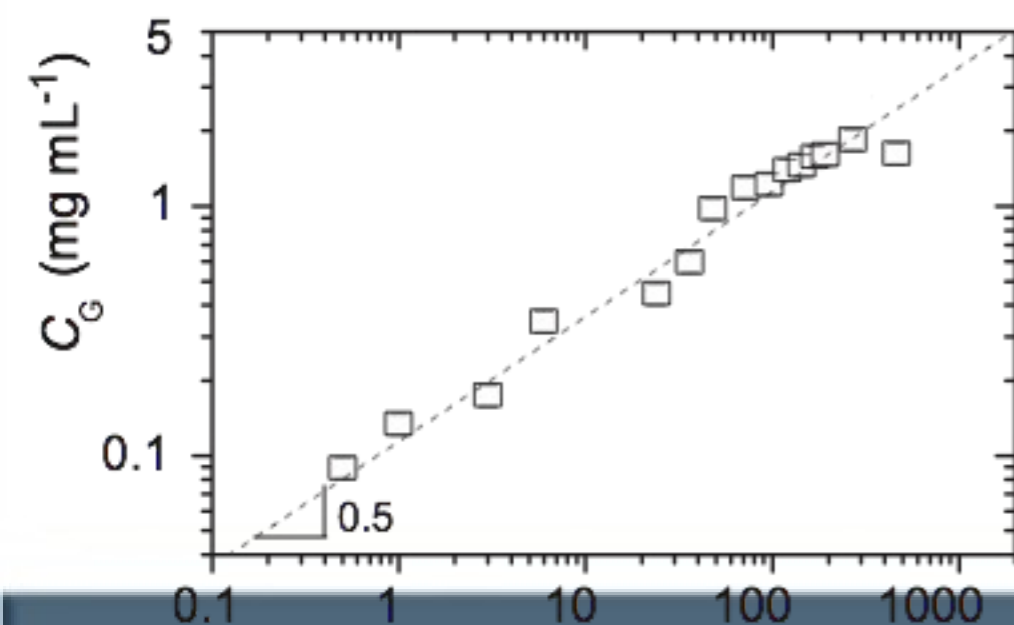
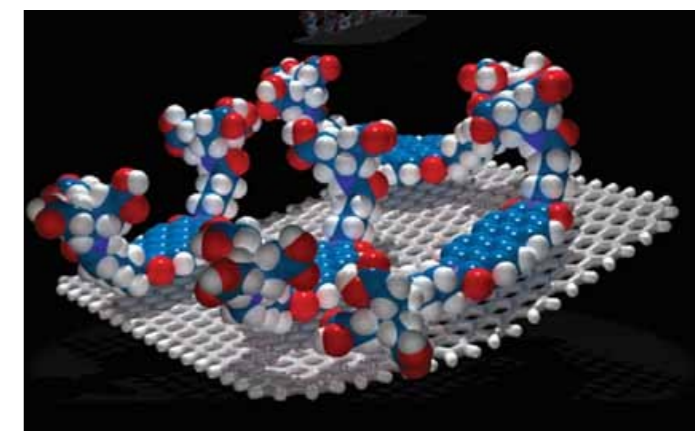


Khan, Small, 2010

Surfactant

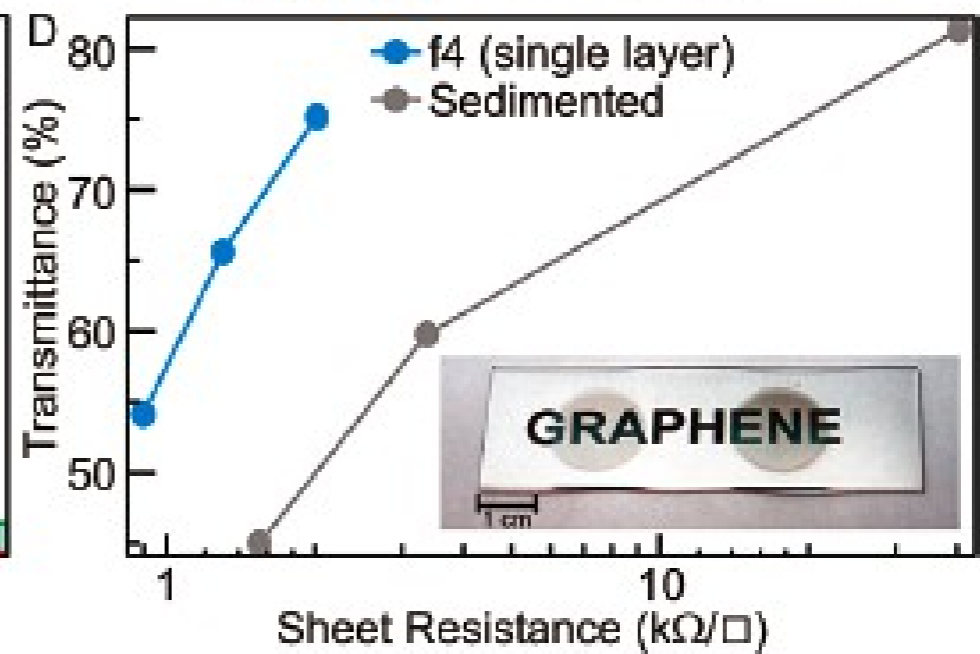
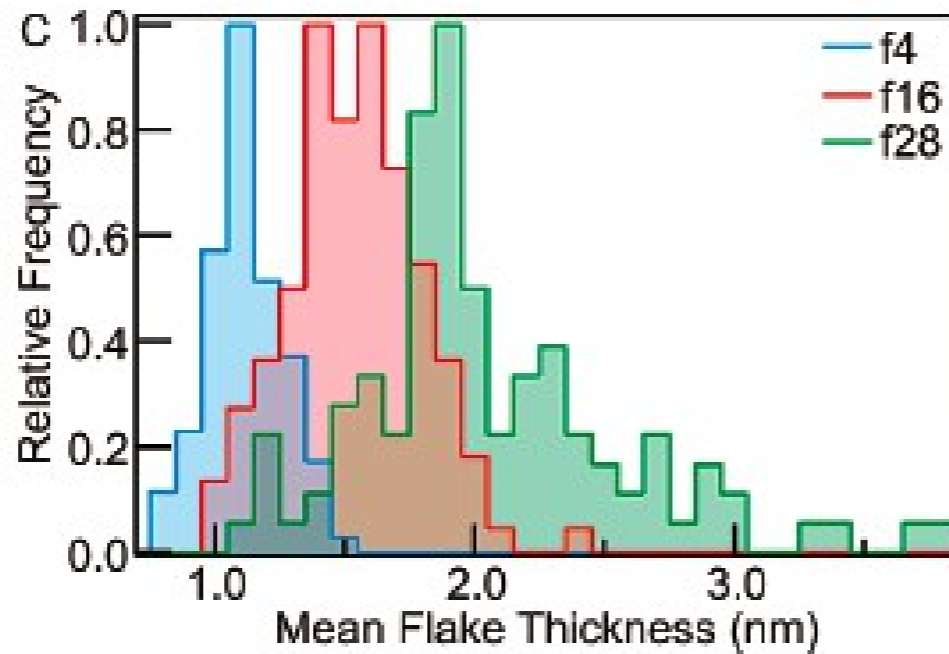
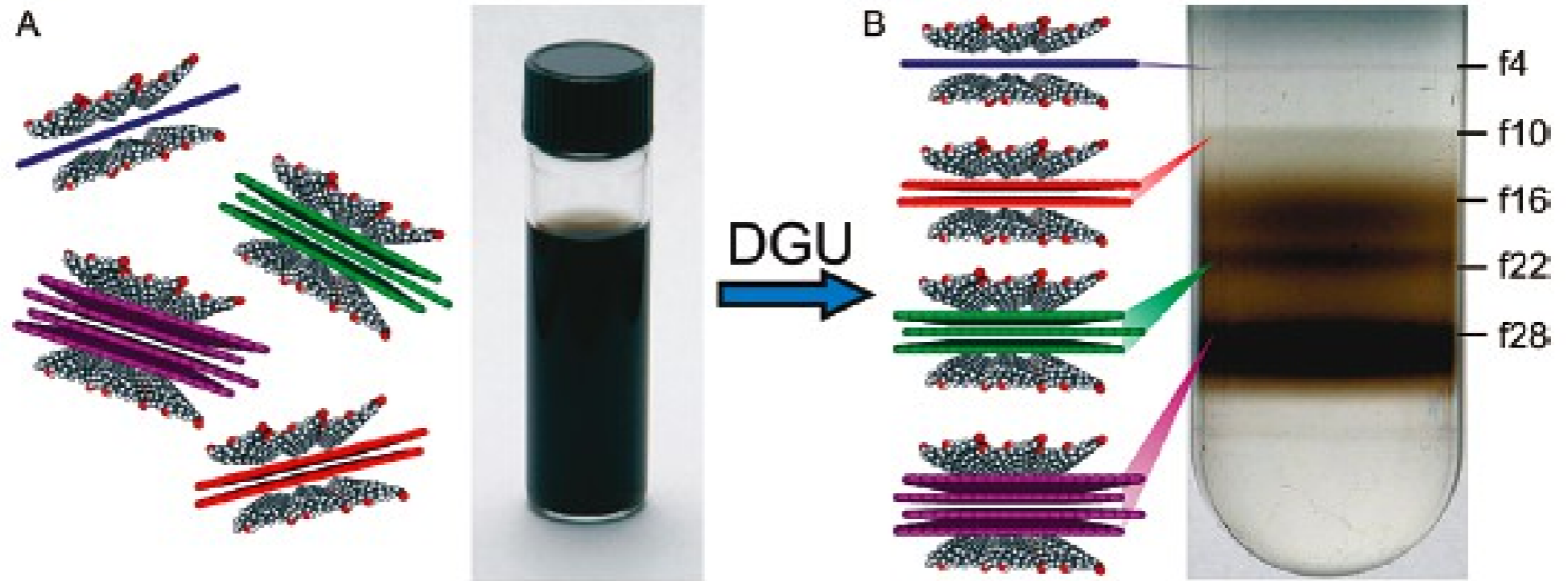


Englert, Adv. Mater., 2009



- > Strong sonication
- > Defects
- > Decrease size

Graphene from Graphite



Green and Hersam, J. Phys. Chem. Lett. 2010
Green and Hersam, Nanolett, 2009

REDUCTIVE DISSOLUTION

Little bit of background...

REDUCTIVE DISSOLUTION

In the case of carbon nanotubes:

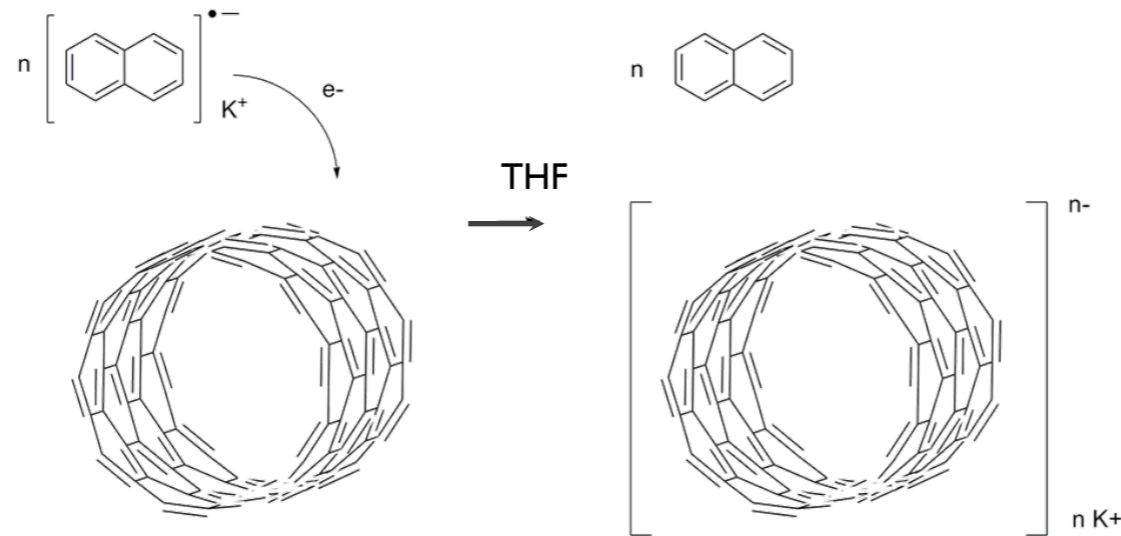
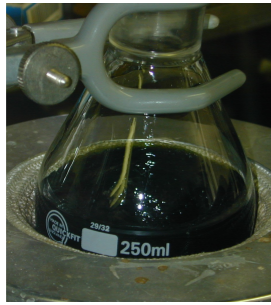
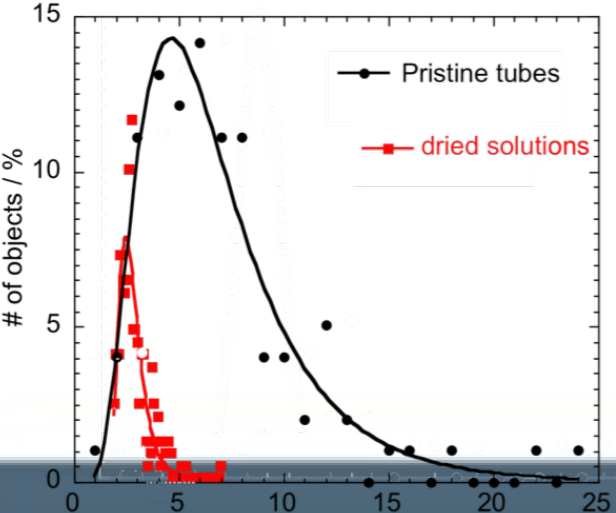
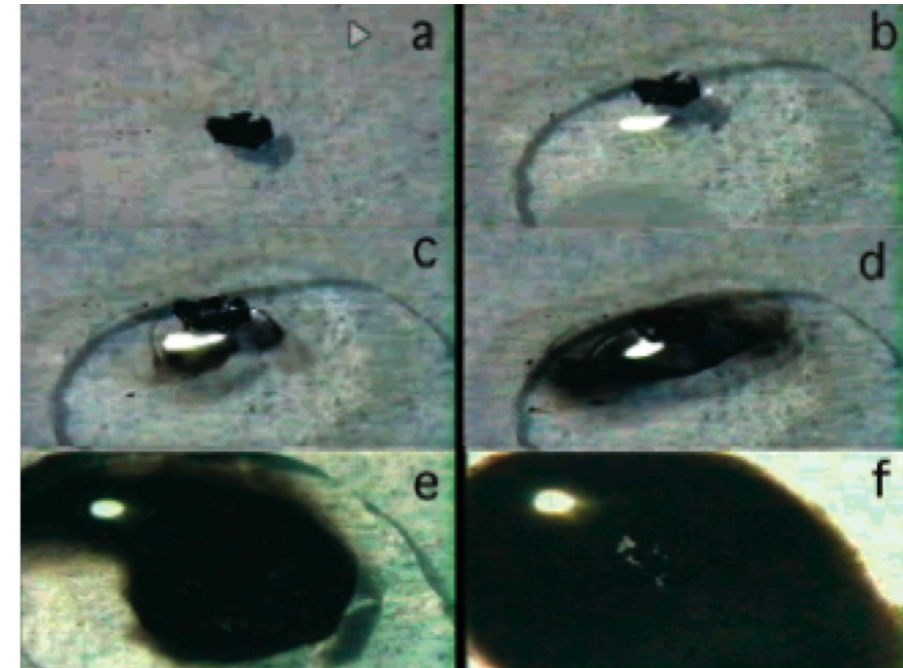
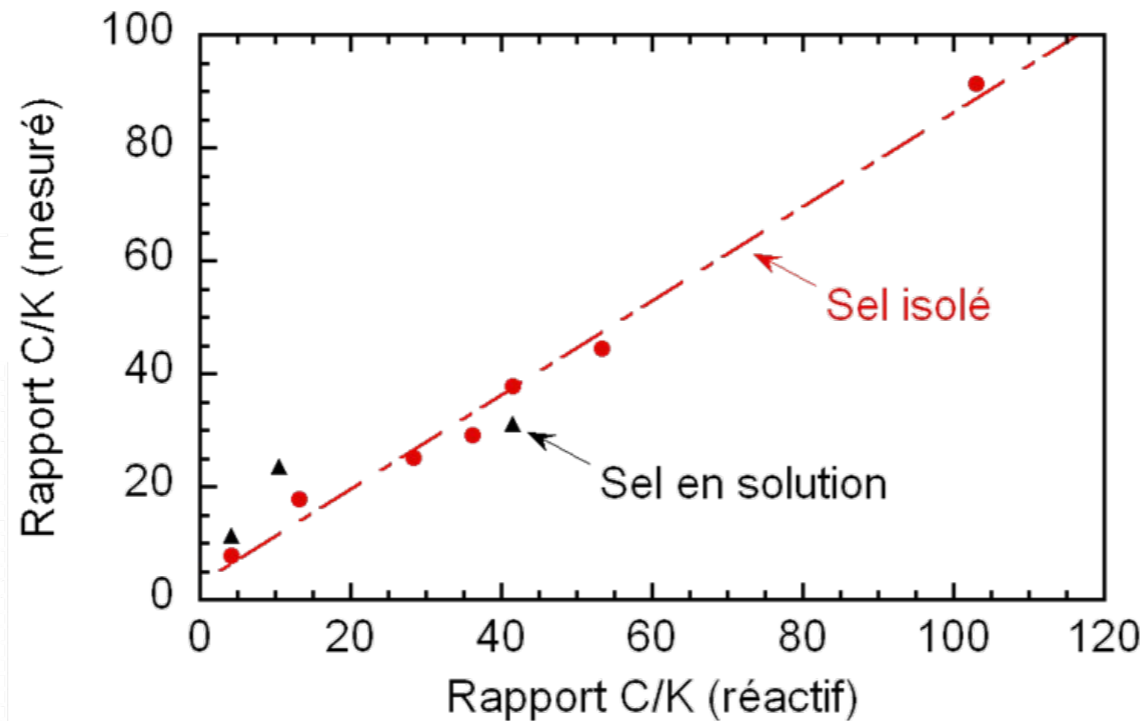


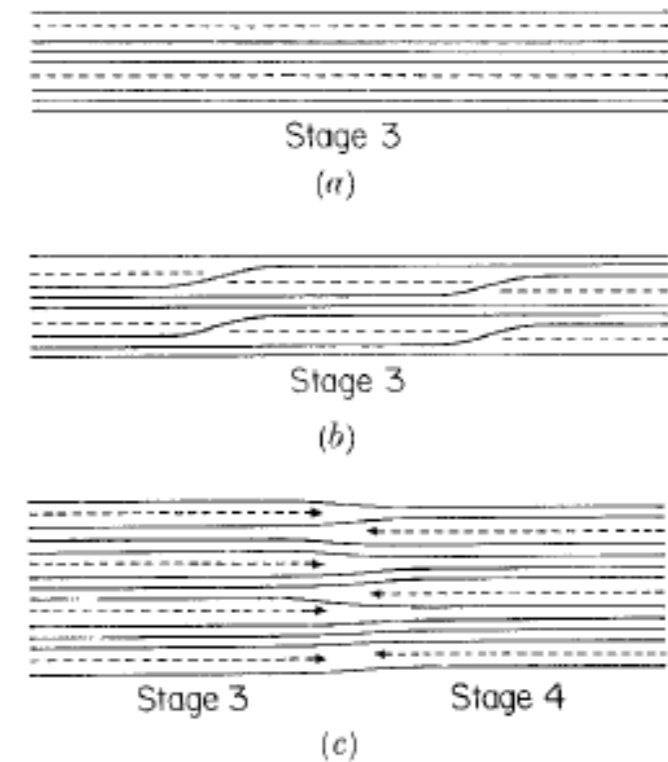
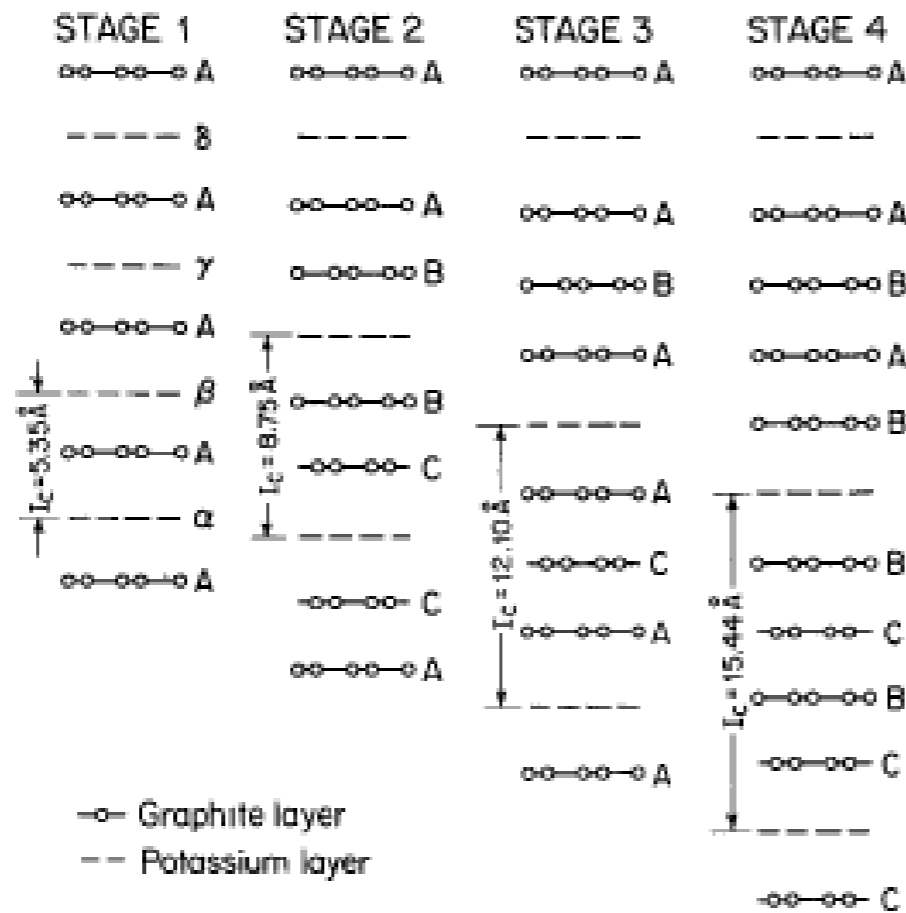
Photo H. Saadaoui



Pénicaud, JACS, 2004

Voiry, J. Mat. Chem., 2010

Graphene from Graphite Intercalation Compounds

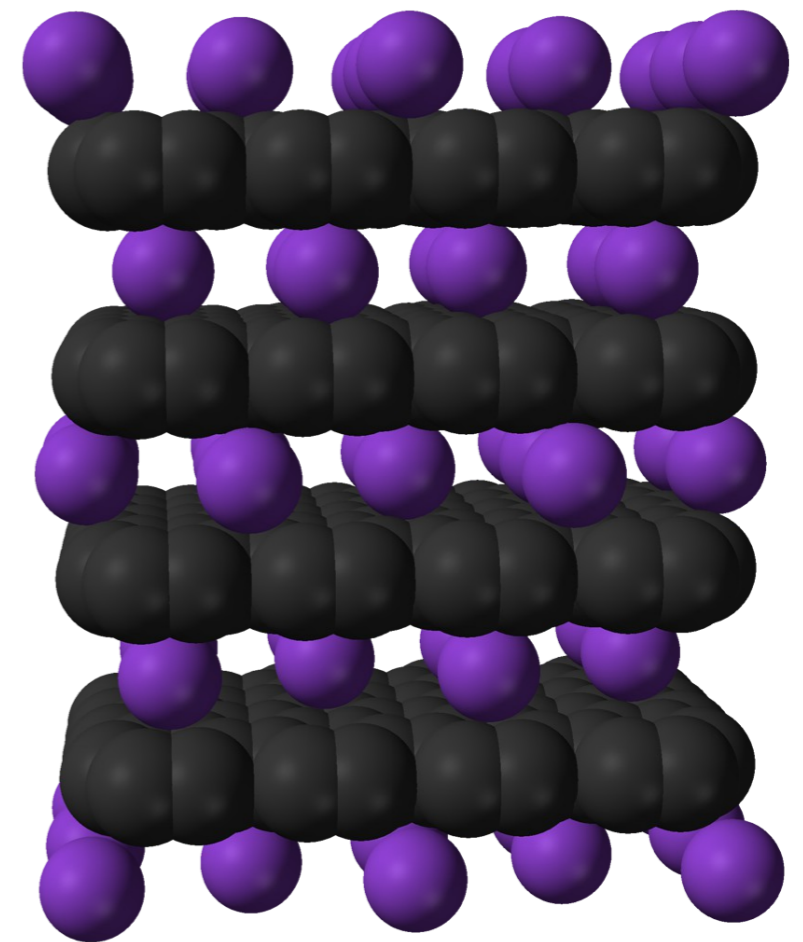


Daumas & Hérol

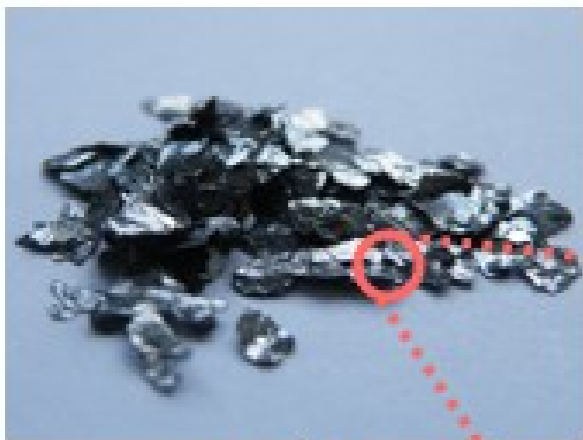
Graphite Intercalation Compounds



M. Dumas©Marabout



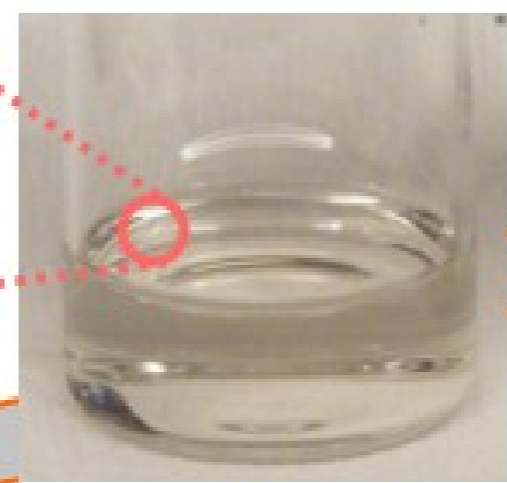
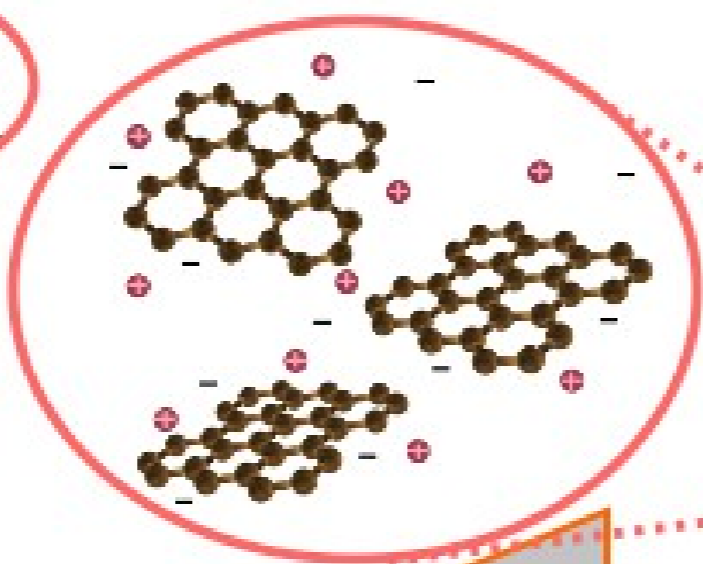
(Image from Wikimedia Commons)
(data from P. Lagrange et al.,
Annales de chimie, 1978)



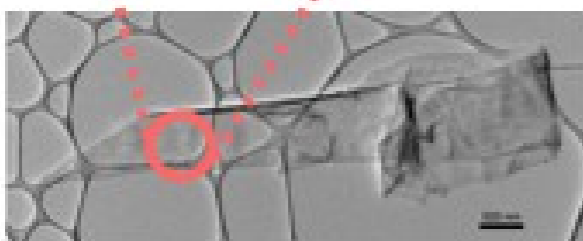
graphite



Fredenhagen & Cadenbach, *Z. Anorg. Allg. Chem.* **1926**



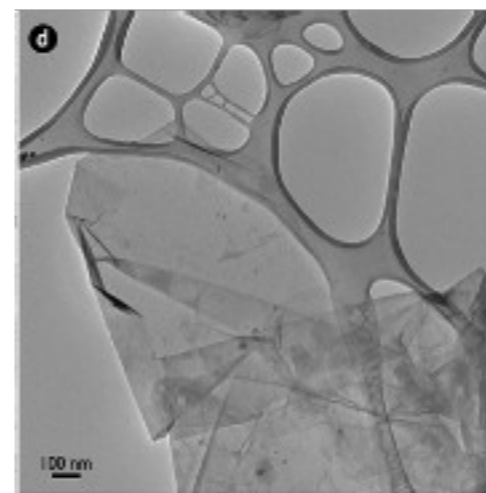
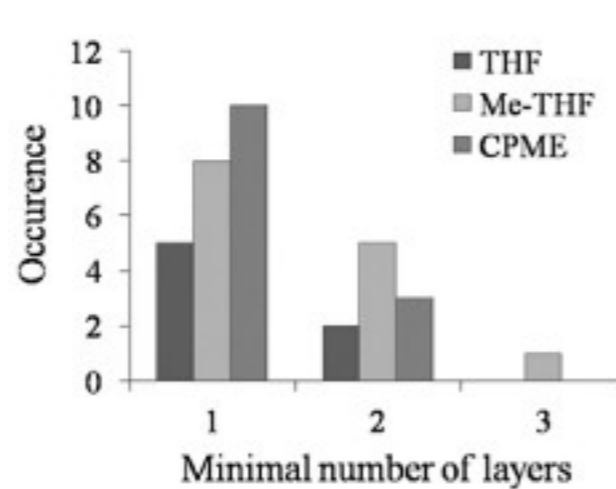
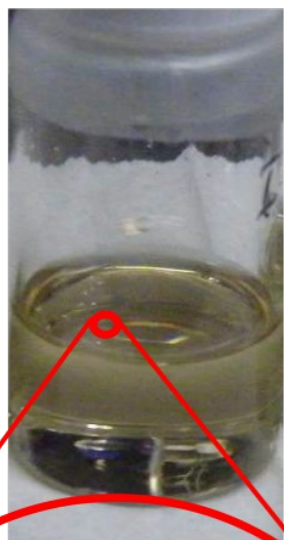
graphenide solution
(Up to 0.7 mg/ml)



graphene

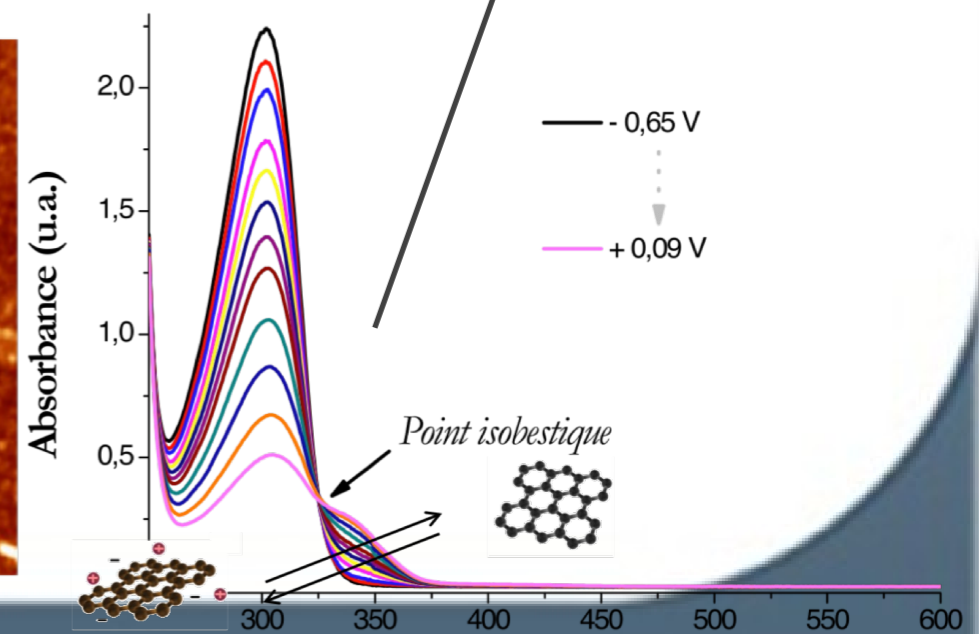
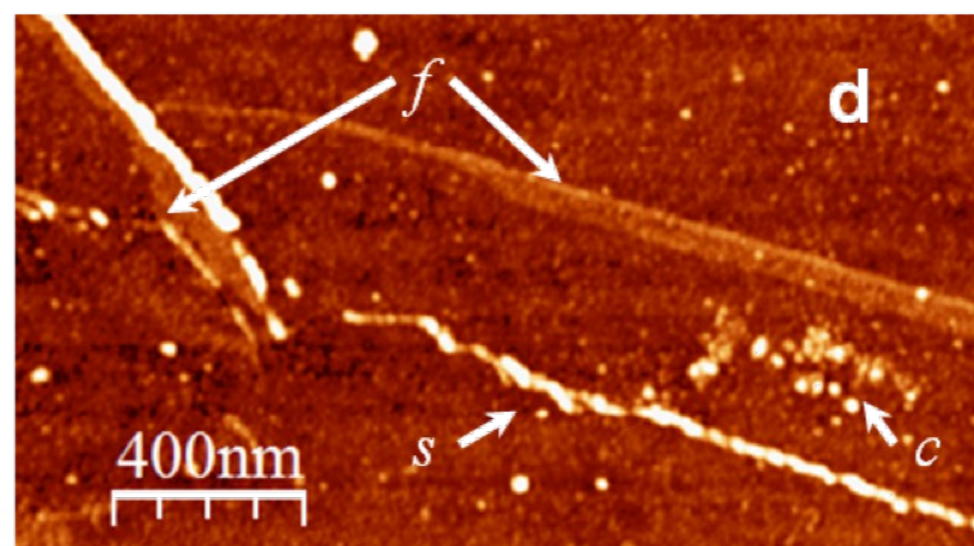
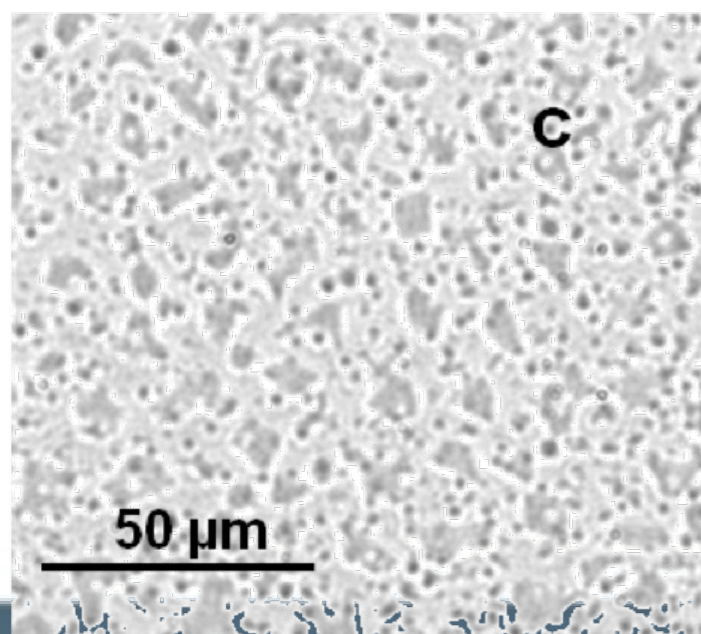
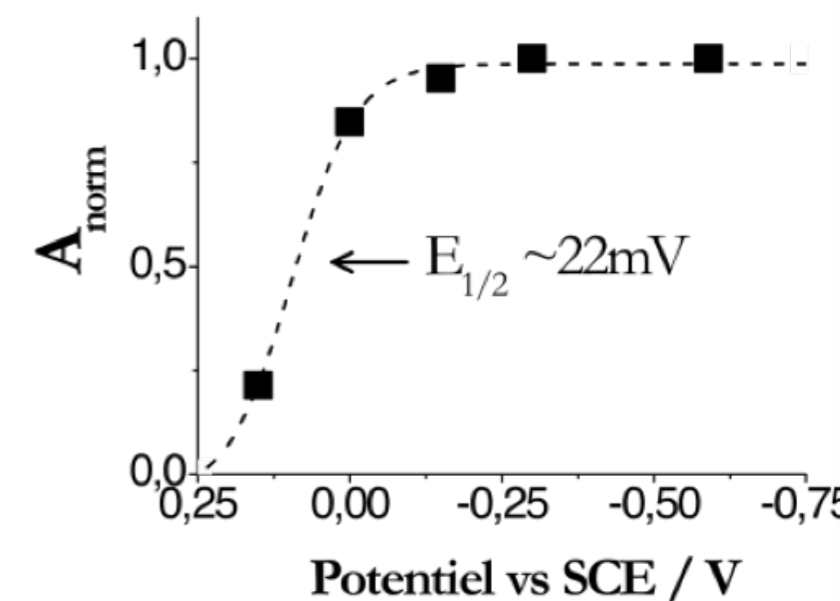
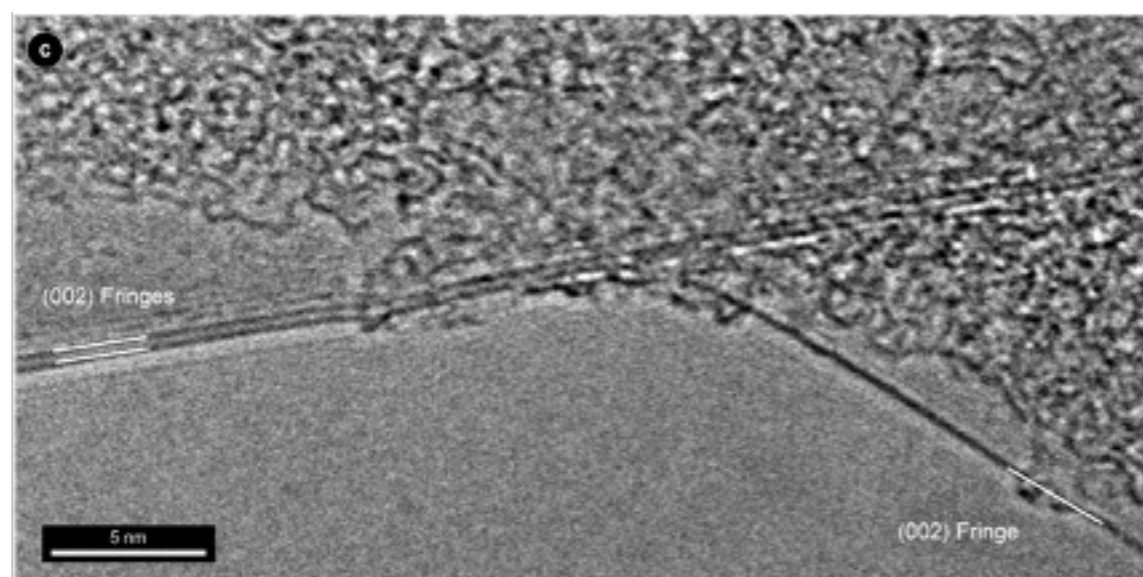
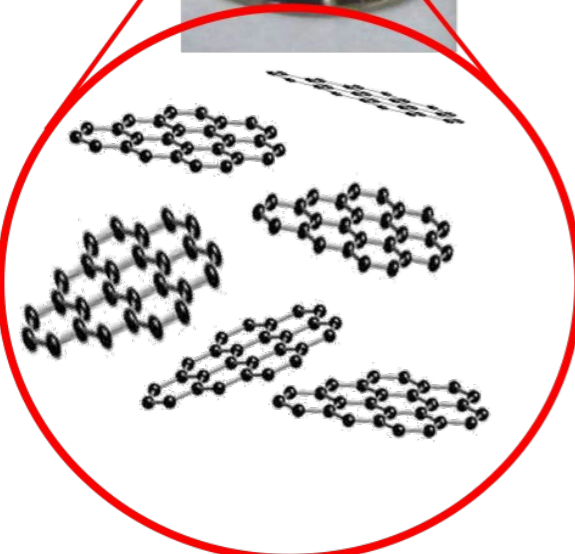
Vallés et al., Patent, 2007, Vallés et al., *J. Am. Chem. Soc.*, 2008, Catheline et al. *Chem. Commun*, 2011
Catheline et al., *Soft Matter*, 2012, Pénicaud & Drummond, *Accounts of Chemical Research*, 2012

Exfoliation of graphene



Catheline, Chem. Commun, 2011

Catheline, Soft Matter, 2012



Selling graphene by the ton

Michael Segal

Small start-up companies are making large volumes of graphene, the world's thinnest material, for applications such as composites and electrodes.

Graphene is, by any measure, a story of extremes. Consisting of a few sheets of carbon atoms, it is the strongest material ever measured², it has a thermal conductivity more than double that of diamond² and has a charge mobility that is among the highest of any semiconductor³. But just as remarkable has been the speed at which this material has moved from the research laboratory to the marketplace.

The exceptional electrical properties of graphene were discovered only five years ago in experiments that involved manually peeling samples weighing picograms from bulk graphite⁴. Today the total production output of various flavours of graphene exceeds 15 tons per year, produced in more than 40,000 square feet of facilities, and this is set to increase to more than 200 tons per year within the next year or two. "We actually have a company asking the question right now, 'When will I get 5 tons?'" says Bor Jang, the chief executive officer of Angstrom Materials, which is based in Dayton, Ohio. "We can't produce enough."

"The graphene is already there — all you have to do is peel it off."

Bor Jang.



A shipment of Vor-x — a functionalized graphene — ready to be sent to a development partner.

The use of acids can lead to oxidation of the graphite, and Vorbeck has licensed an approach developed at Princeton University^{5,6} to convert graphite oxide into

manufactured will not end up in advanced devices such as transistors. Instead, most applications fall into two broad and not always exclusive categories: composites and electrodes. Composites consist of graphene dispersed into a host matrix, and take on some of the properties of graphene as a result. These may include strength, stiffness, and electrical or thermal conductivity. For example, some containers used to store and transport volatile fuels are made from polymer or rubber composites. Incorporating graphene would increase their stiffness by an order of magnitude, and allow them to conduct electrical currents, reducing the probability of electrical discharges and increasing safety.

Similarly, packaging made from a graphene composite could shield electronics from radio frequency signals, conduct current to external devices or even store information. Vorbeck is focusing on printable conductive inks (a distinct class of composites), for applications in printed electronics such as radio-frequency identification tags, low-resolution displays and backlights, sensors, flexible connectors and packaging. Graphene makes these inks both highly conductive and easy to process.

Graphene-based electrodes take advantage of the material's electrical conductivity, transparency and surface area. One attractive target is the replacement of indium tin oxide.

What is graphene?

Although the International Union of Pure and Applied Chemistry (IUPAC) defines

graphene as **a single layer of carbon atoms**

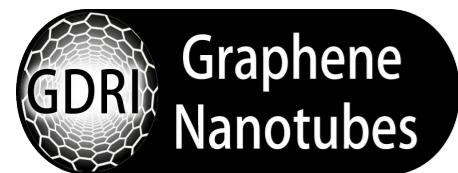
researchers routinely talk of bilayer, trilayer

bilayer, trilayer and multilayer graphene

the definition further still, to include any

any highly exfoliated graphite product

uses predominantly single-layer graphene



Chem nTubes 2012

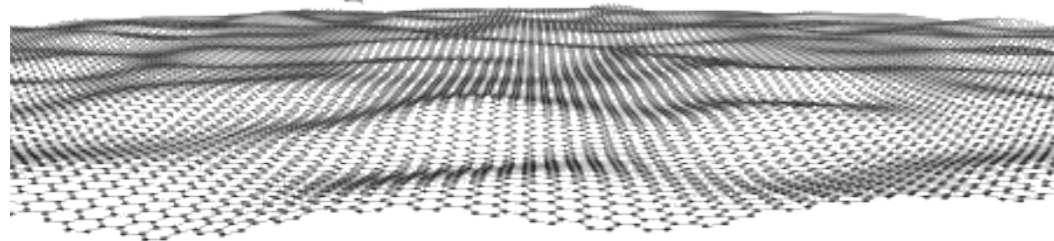


The “Nanotube and Graphene” team of CRPP

Amélie Catheline, Damien Voiry, Sandra Enriquez, Yu Wang, Fabienne Dragin, Elisa Del Canto, Kai Huang, Carlos Drummond, Cécile Zakri (CRPP)
Eric Anglaret (Montpellier), Pascal Puech, Marc Monthieux (Toulouse)
Luca Ortolani, Vittorio Morandi (CNR Bologna)
F. Paolucci, M. Iurlo, M. Marcaccio, S. Rapino, G. Valenti (Bologna)



ChemOnTubes 2014



March 30th - April 3rd, 2014

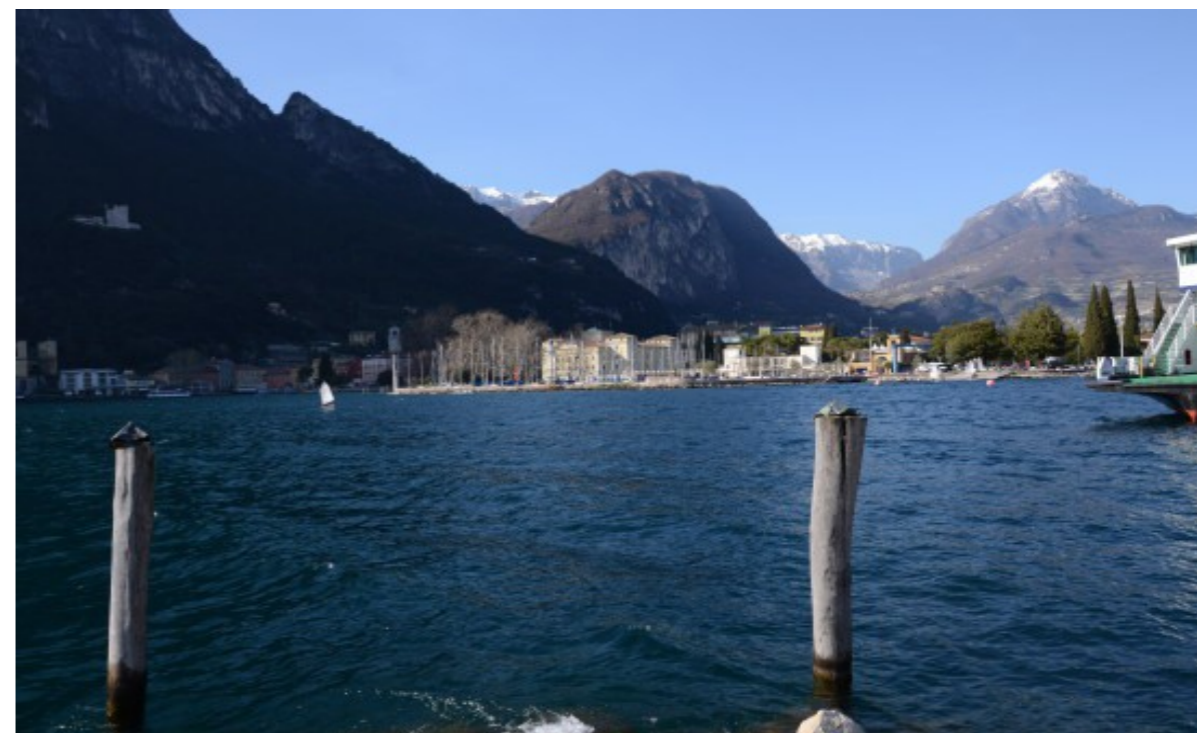
Riva del Garda, Italy

**International Meeting on the Chemistry of
Graphene and Carbon Nanotubes.**

Scope

All sessions will be devoted to both graphene and carbon nanotubes

- Functionalization, dispersion, sorting
- Electrochemistry, devices
- Composites, foams, coatings
- Energy storage, conversion, harvesting
- Nanomedicine, biomaterials
- Functional materials
- Catalysis, filtration, membranes
- Organic electronics



Abstract submission deadline: December 12th, 2013

Registration deadline: January 31, 2014

Contact : ChemOnTubes@crpp-bordeaux.cnrs.fr

Graphene from carbon nanotubes

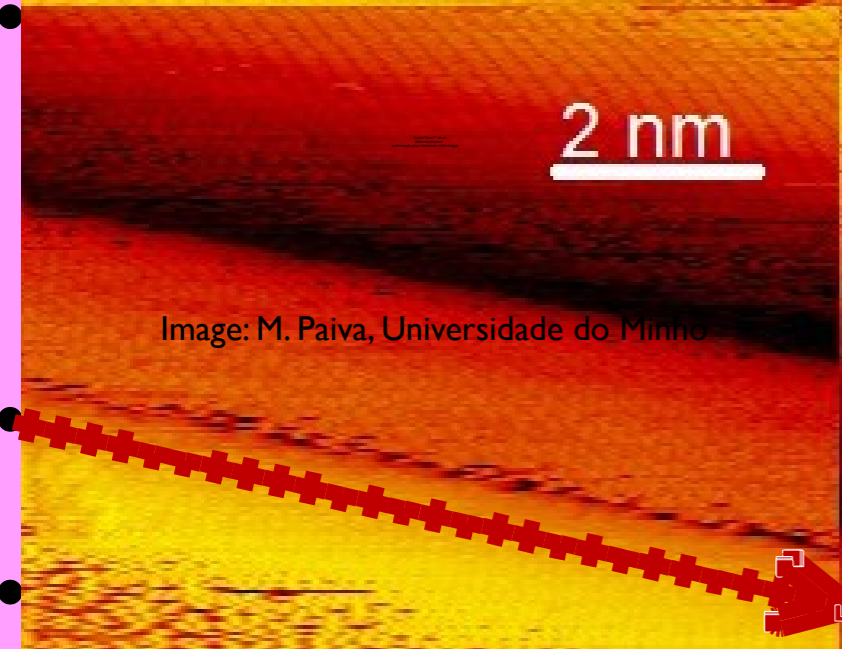


Image: M. Paiva, Universidade do Minho

characterize them?

- Graphene from Graphite
- Graphene from GICs
- Graphene from carbon nanotubes
- Graphene from molecular precursors
- Ton-scale graphene



Industry has extended not only the definition of graphene, but also its history, questioning the claim that it is a new material. “The original synthetic work for graphene goes back into **the late 1800s**” says Hahn. “It is hard to look at graphene and say it is a new material.” This early work focused on intercalated or lamellar compounds of graphite, and by 1970 single layer graphene on nickel substrates, called monolayer graphite, had been isolated^{9,10}.

Graphene from carbon nanotubes

Semantics:
graphene,
graphene
nanoribbons,
nanographene,

● Why chemical
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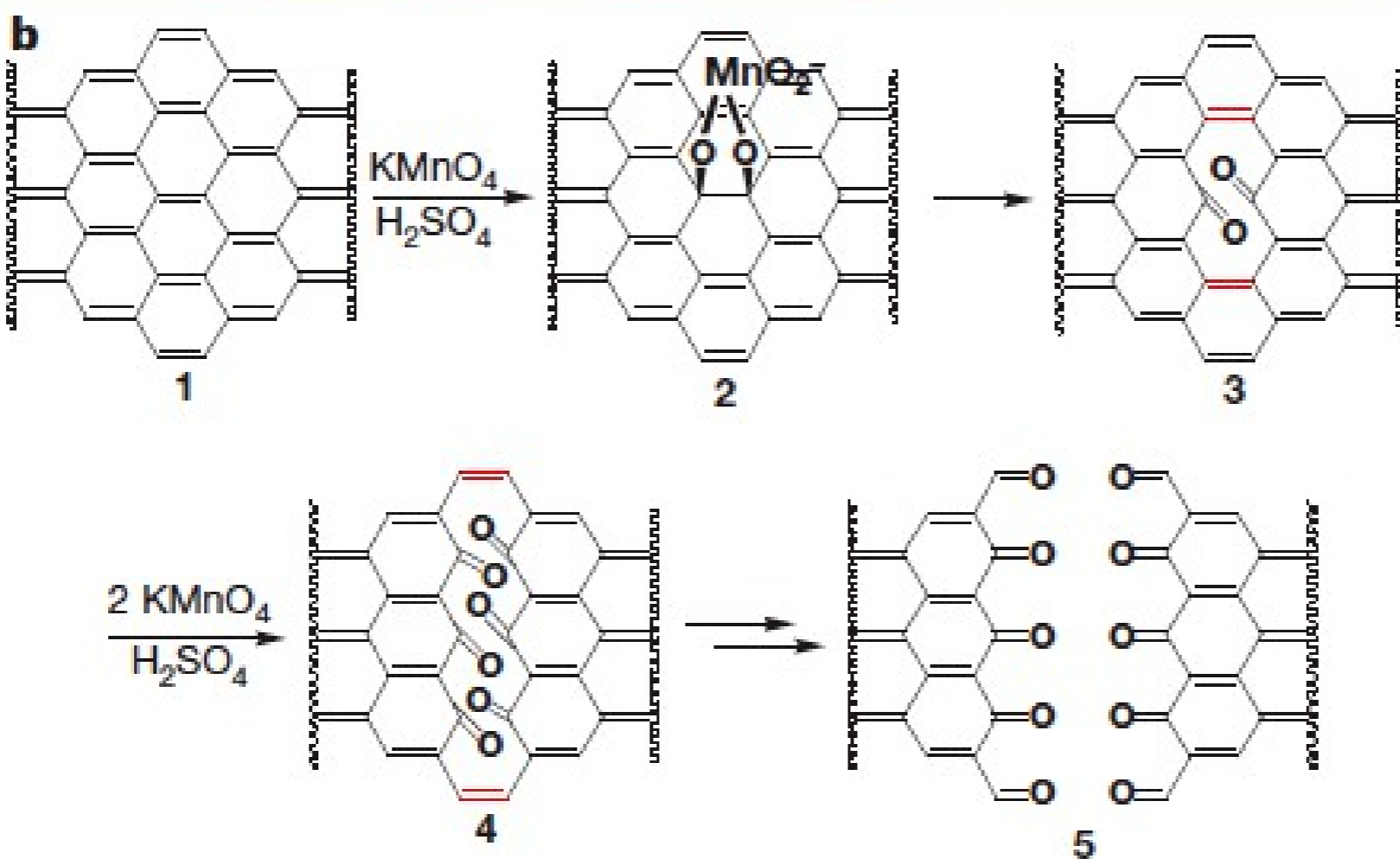
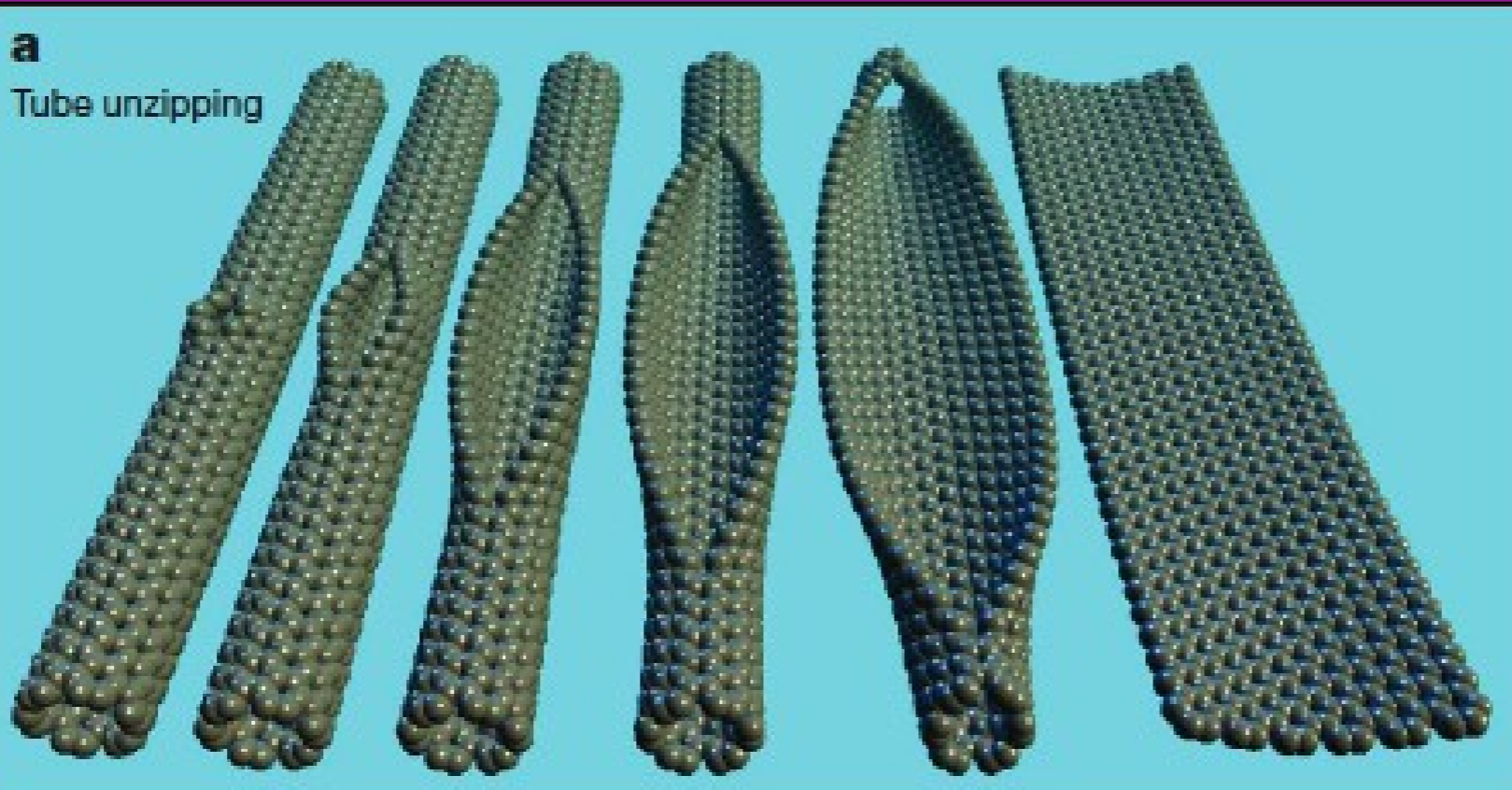
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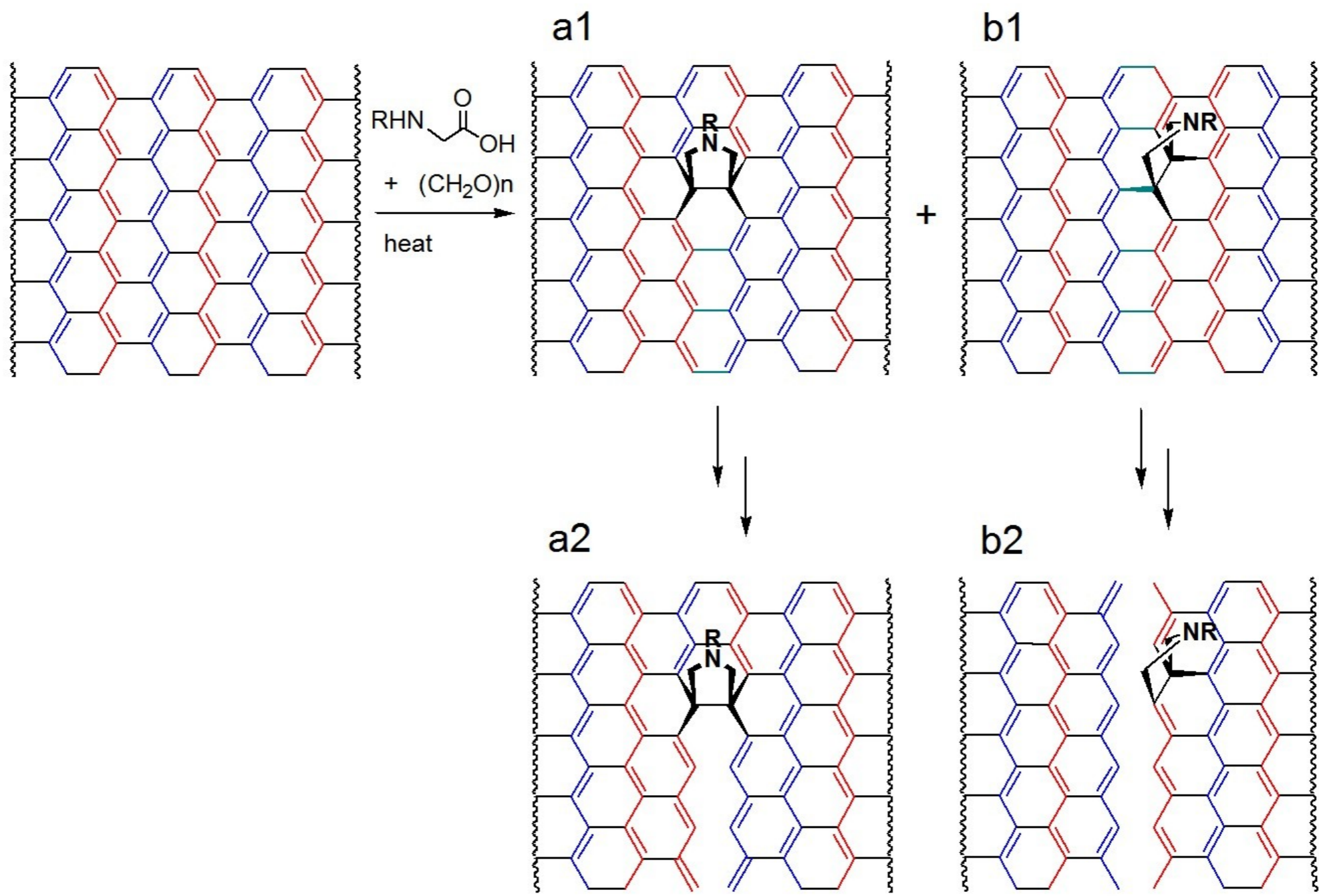
● Ton-scale
graphene
Alain Pénicaud, Cent.



SEM image

Graphene from carbon nanotubes

Unzipping of functionalized CNT to form graphene



Slide from Maria Paiva, Universidade do Minho, Portugal

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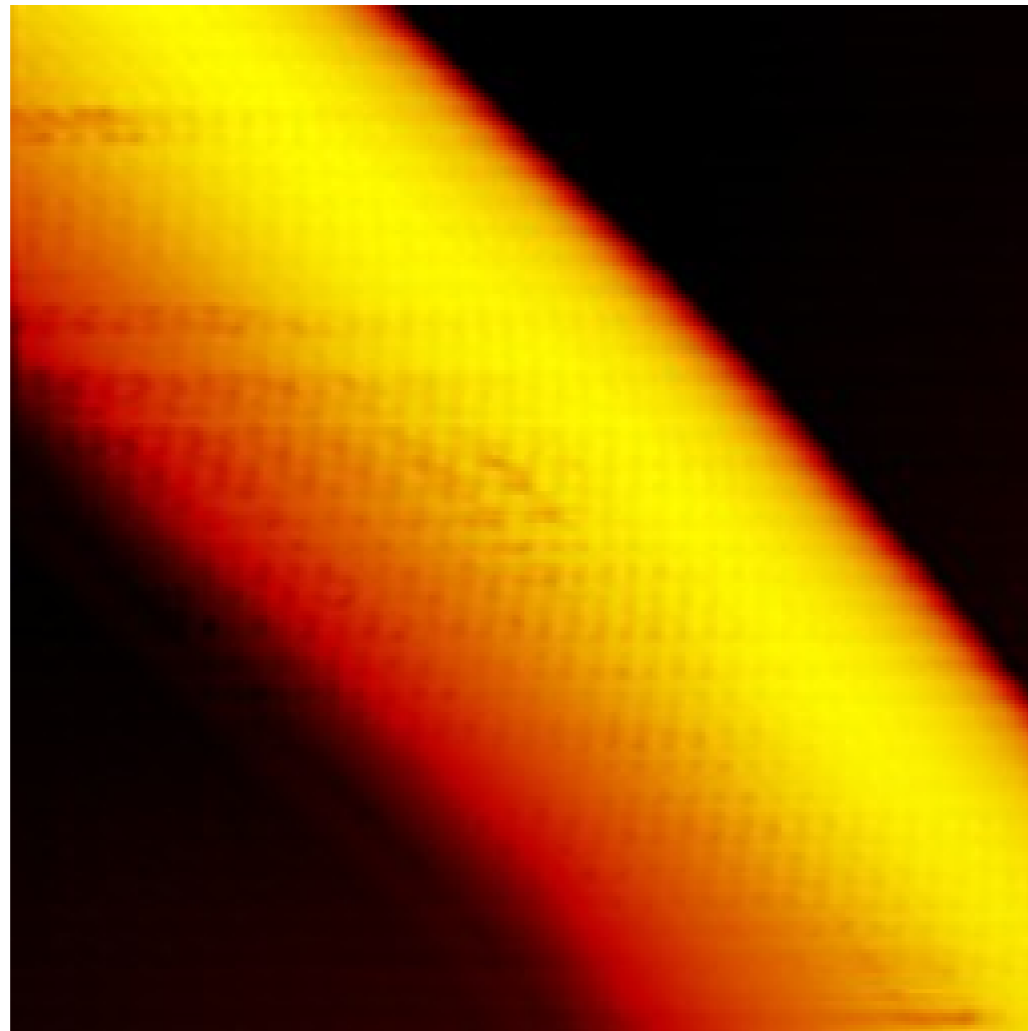
● Ton-scale
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Alain Pénicaut, Centre de Recherche Paul Pascal - CNRS - Université de Bordeaux

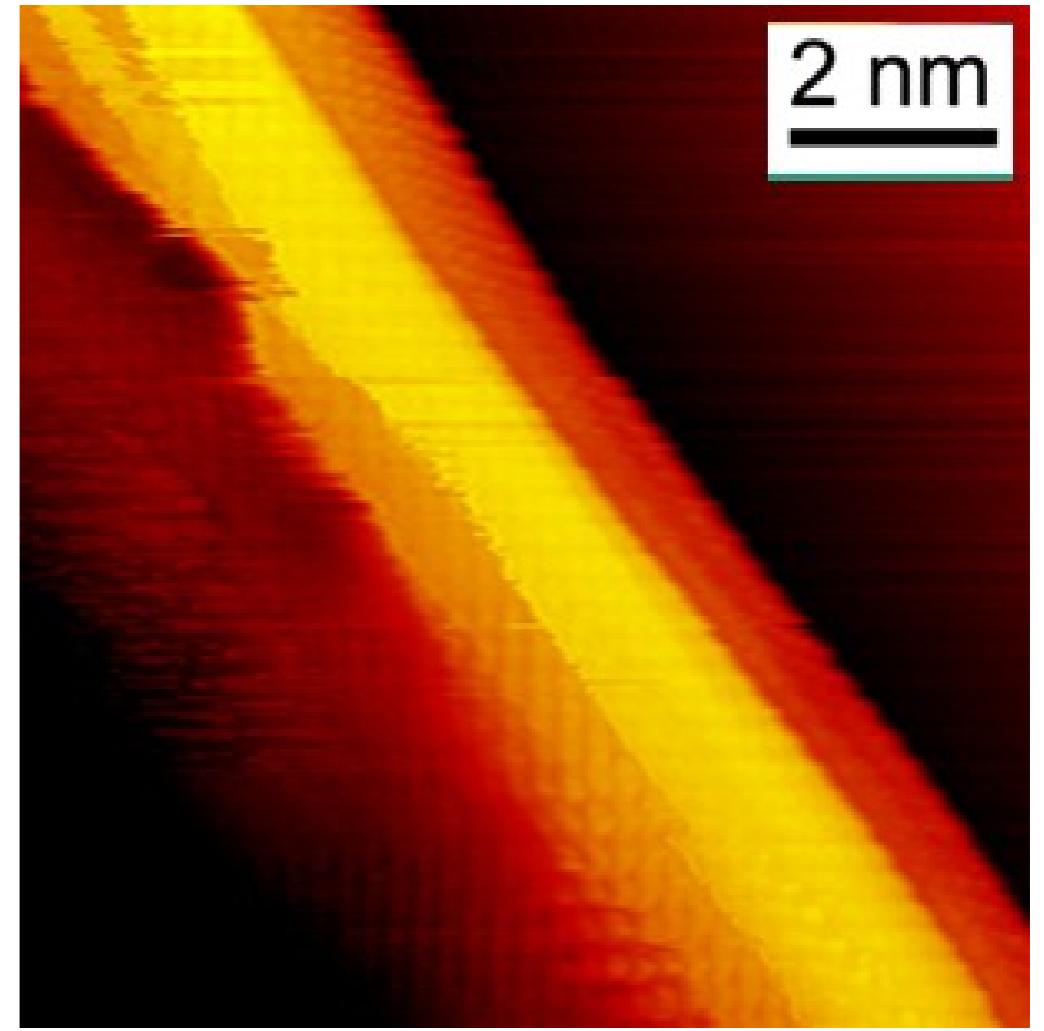
Graphene International School, Cargèse, October 11-22, 2010

Graphene from carbon nanotubes

The 1,3-dipolar cycloaddition to CNT: nanotube unzipping



Non-functionalized nanotube



Unzipping of the outer nanotube layer

M. C. Paiva, W. Xu, M. F. Proença, R. M. Novais, E. Lægsgaard, F. Besenbacher, *Nano Letters*, 2010, 10, 1764-1768

Slide from Maria Paiva, Universidade do Minho, Portugal

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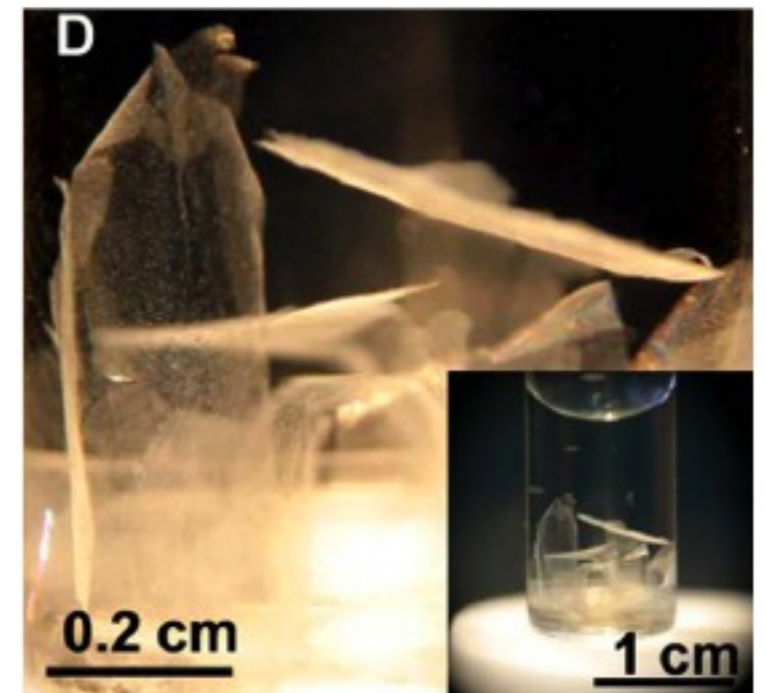
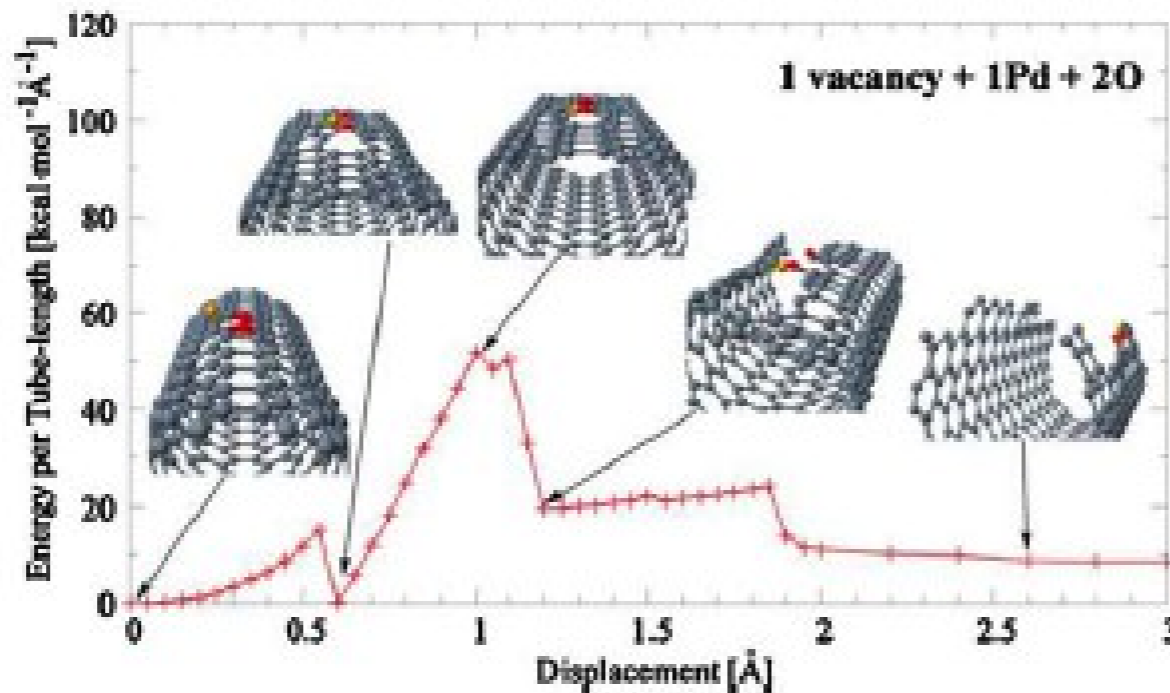
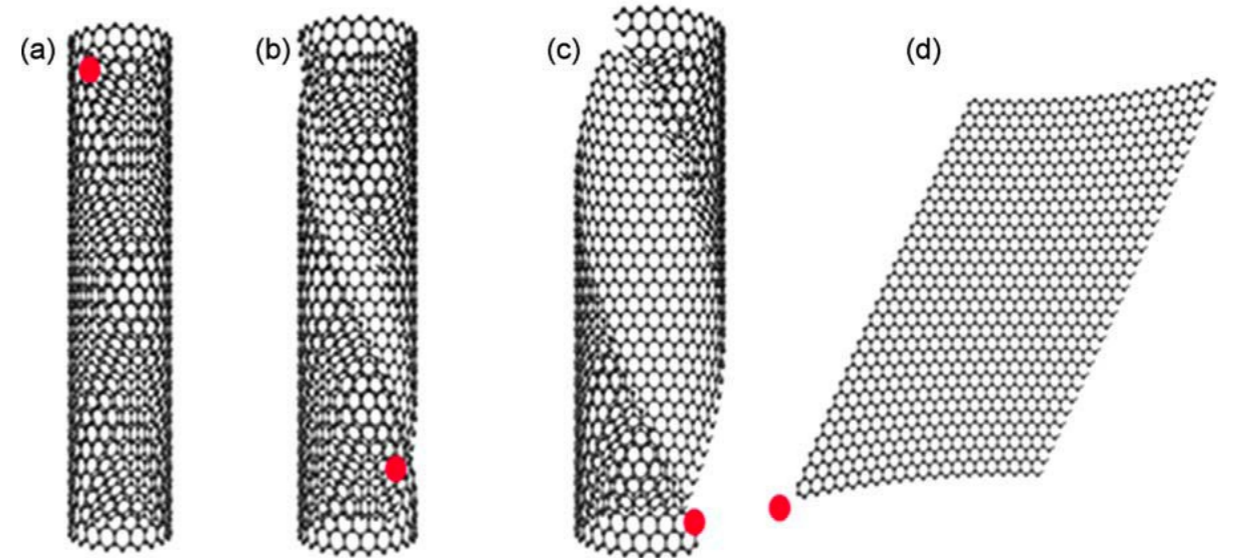
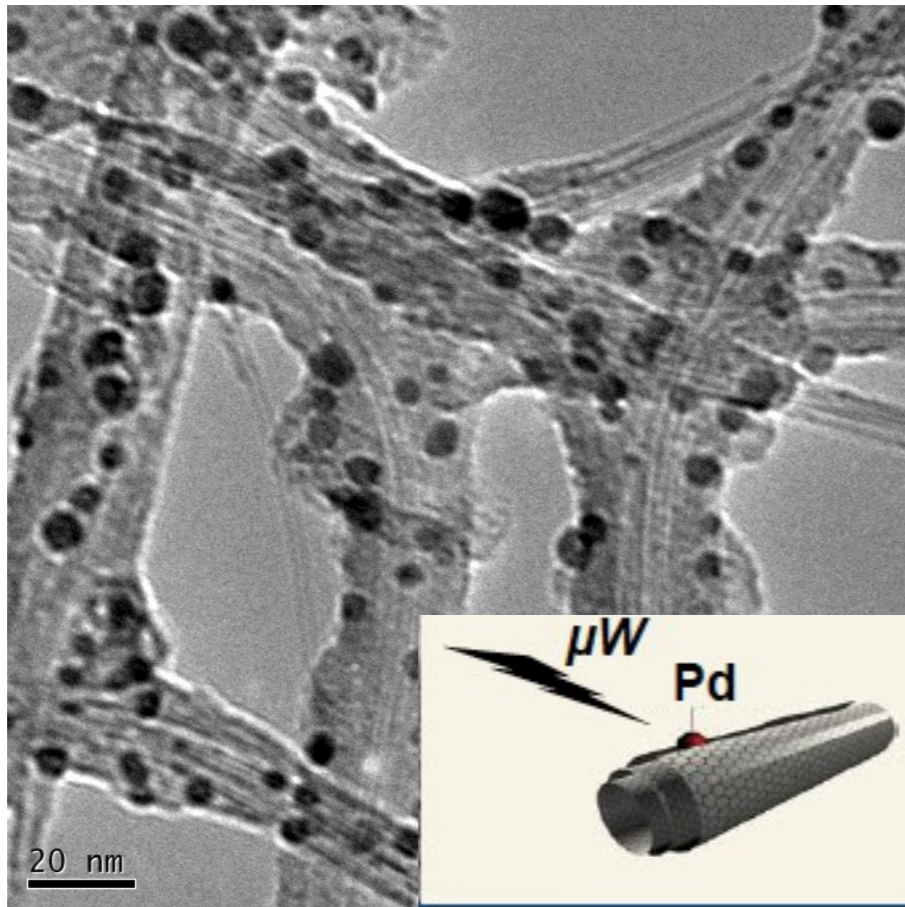
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Alain Pénicaut, Centre de Recherche Paul Pascal - CNRS - Université de Bordeaux



Graphene from carbon nanotubes

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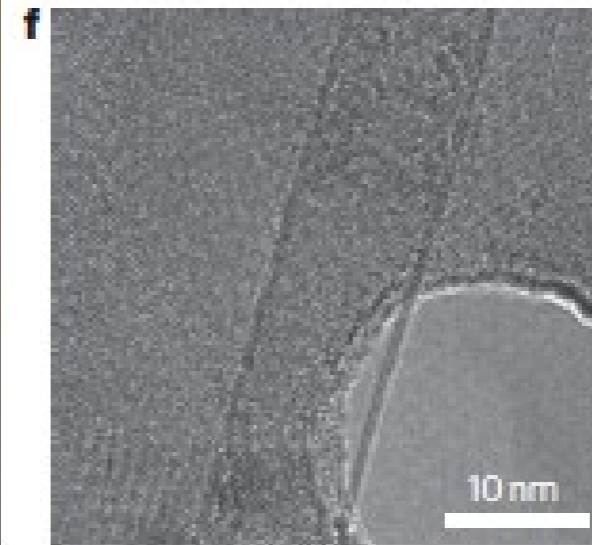
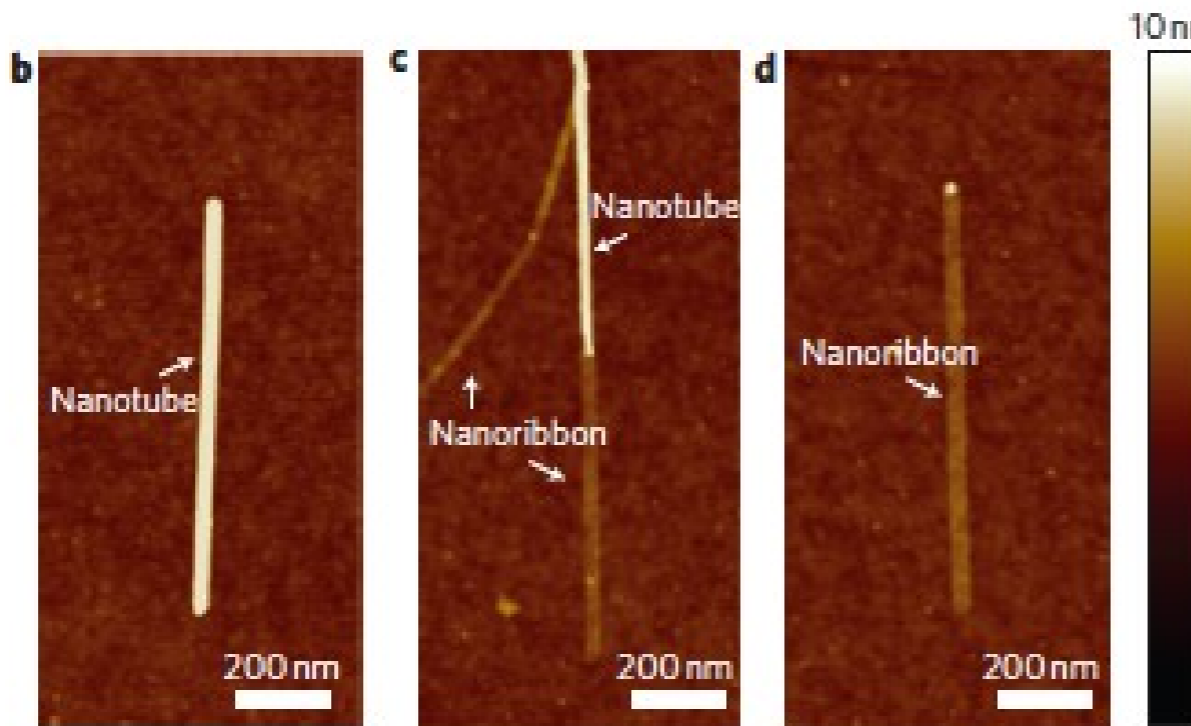
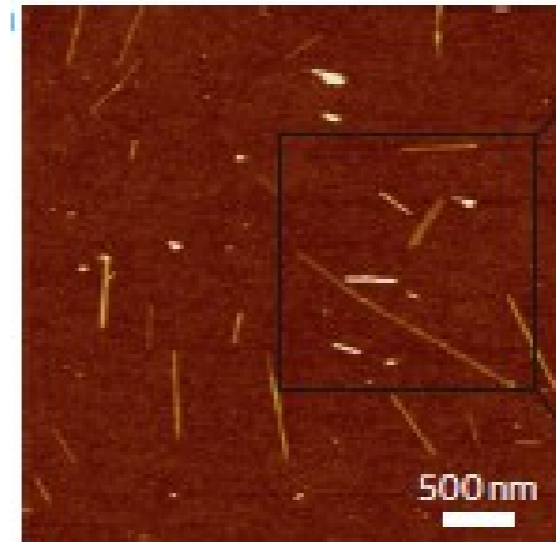
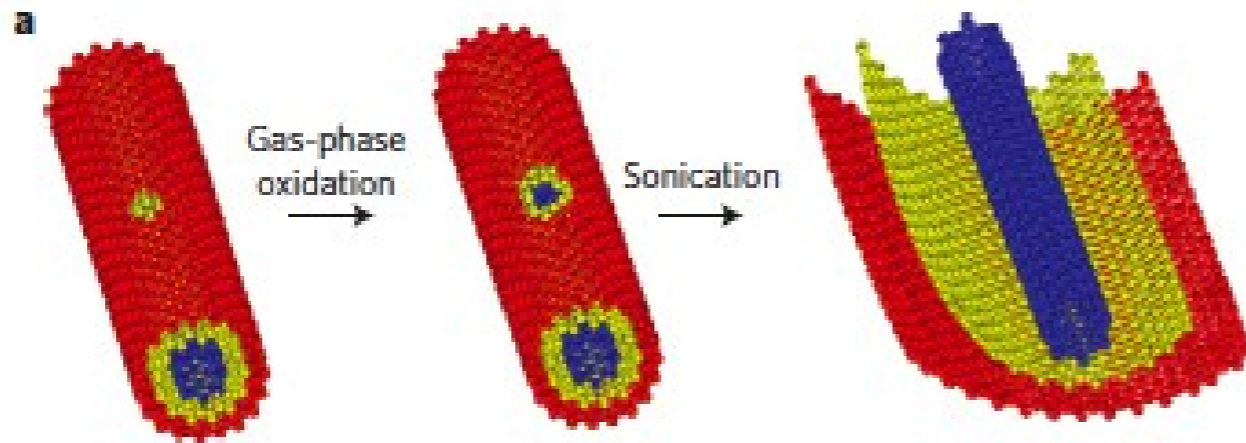
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- Ton-scale graphene

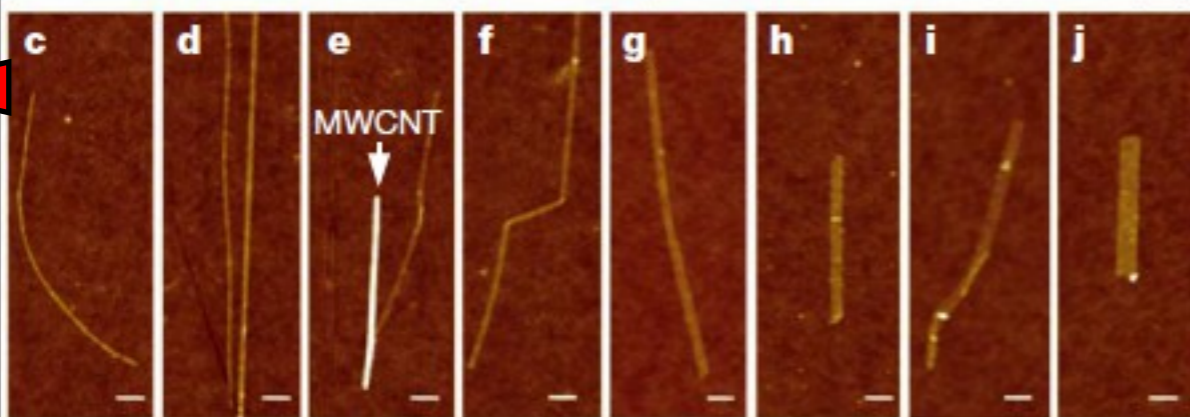
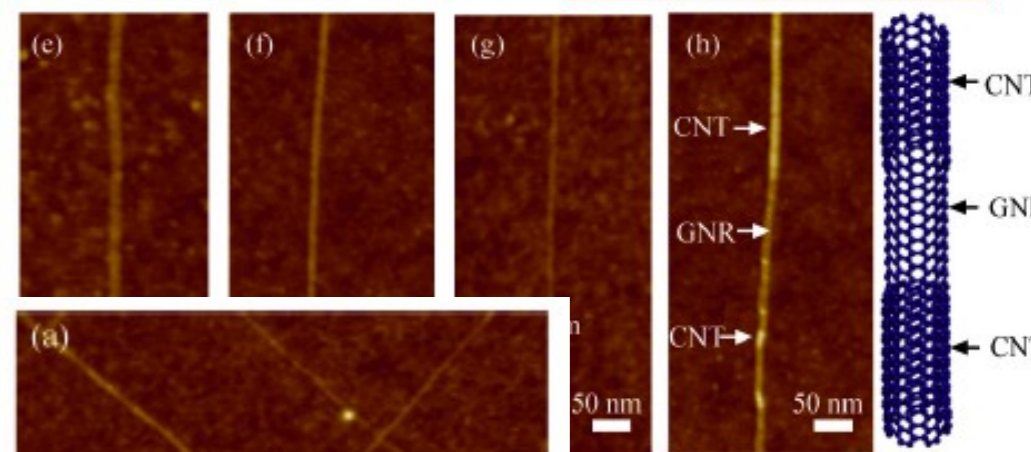
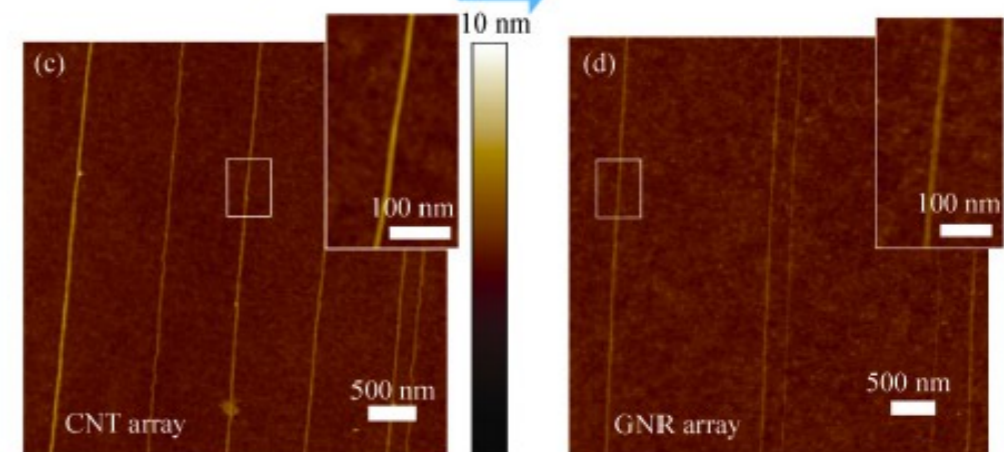
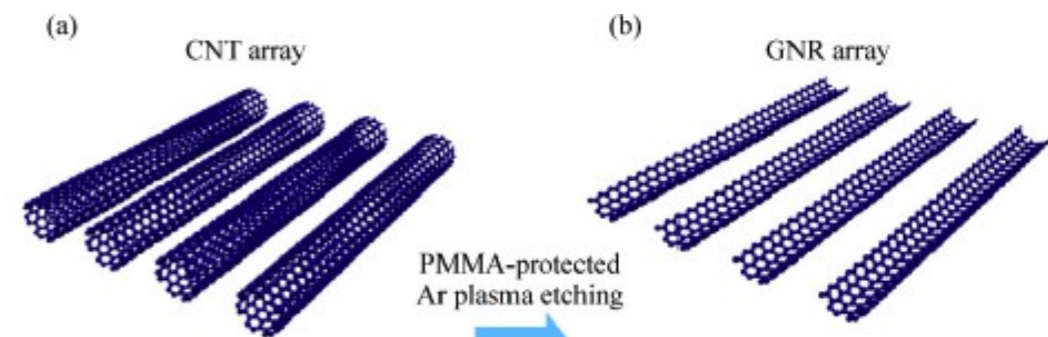
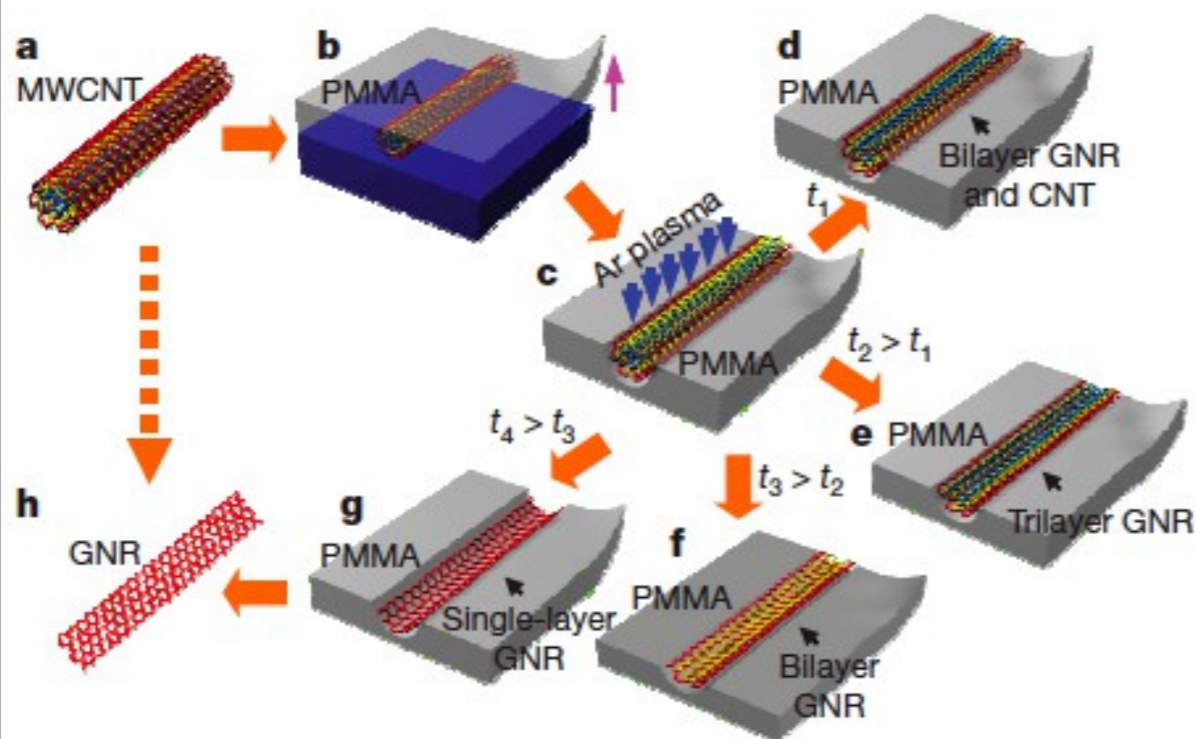
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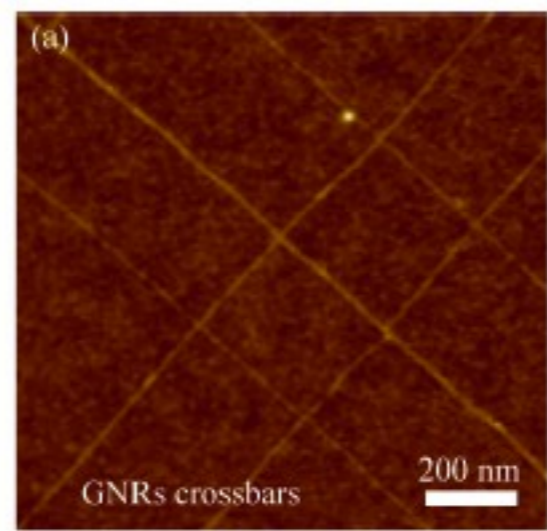
Hongjie Dai et al., Nature Nanotechnol. 2010

Graphene from carbon nanotubes

nanotube unzipping by Ar plasma etching



Hongjie Dai et al. Nature 2009



Hongjie Dai et al. Nanoresearch 2010

● Semantics:
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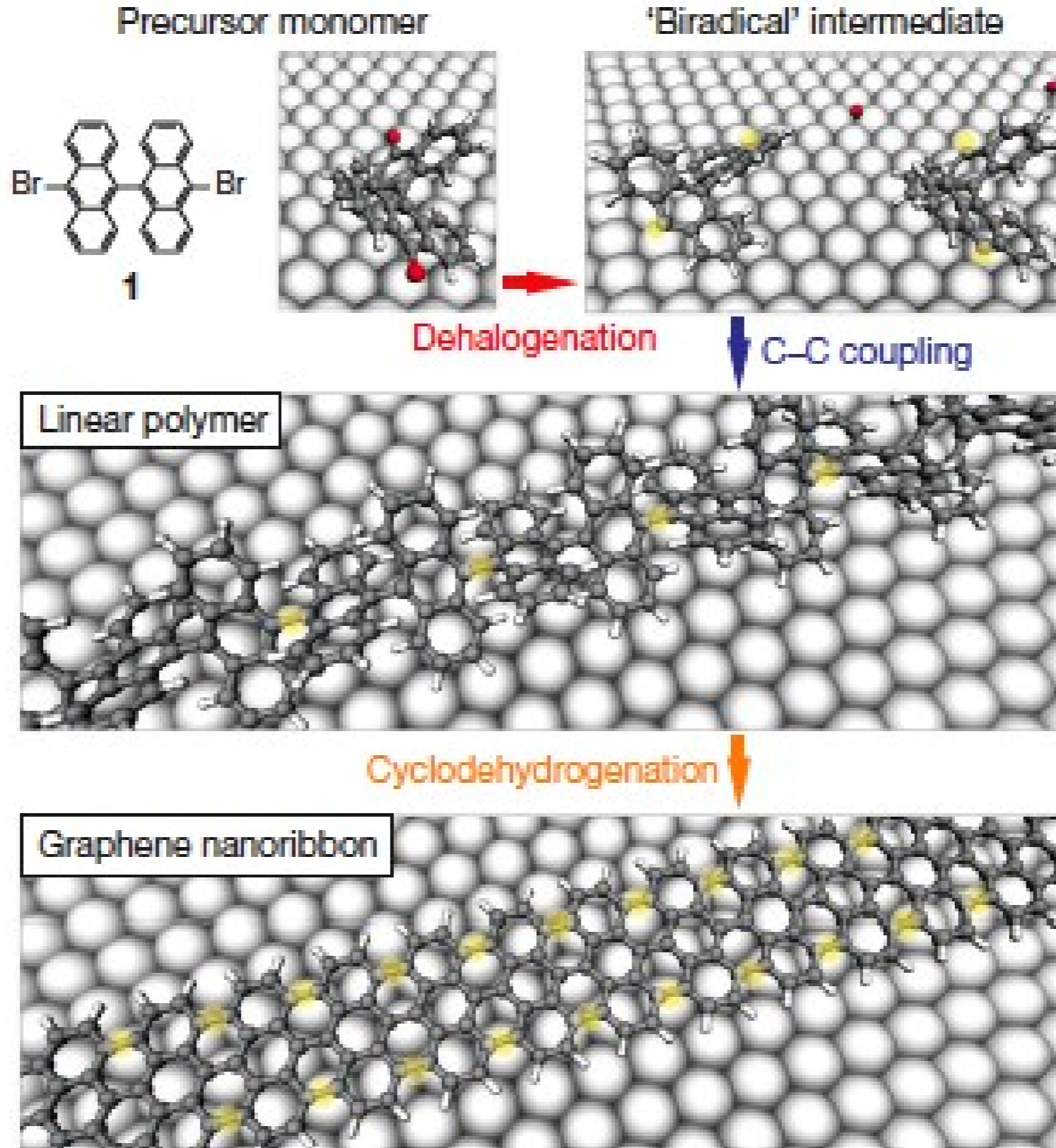
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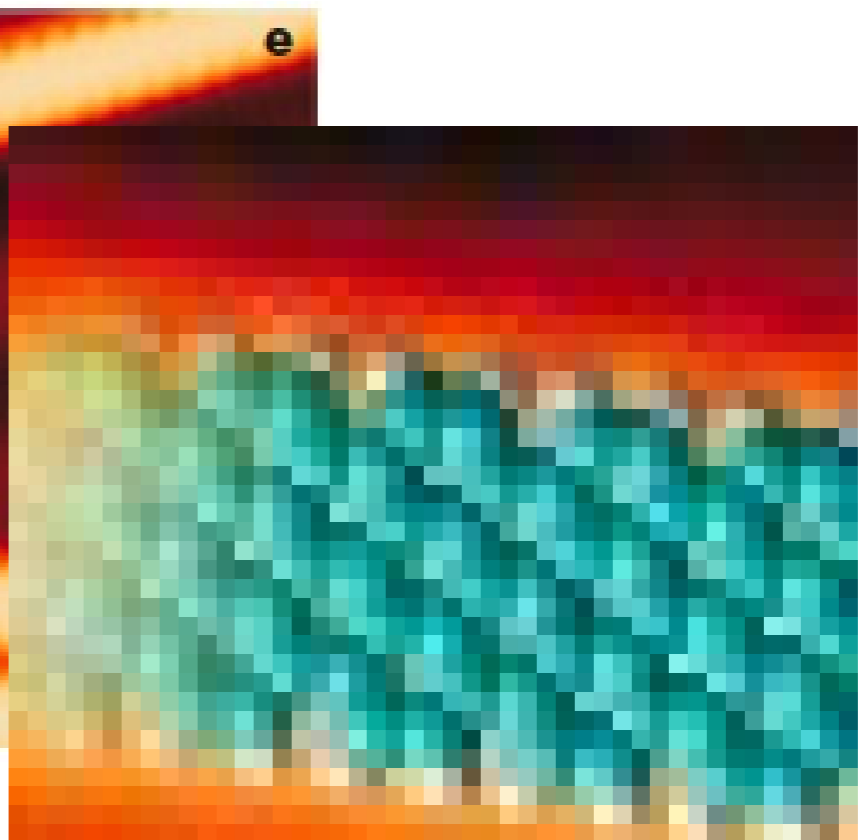
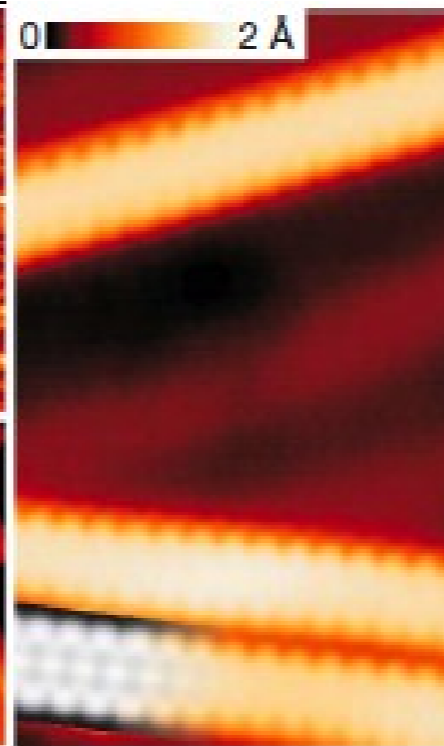
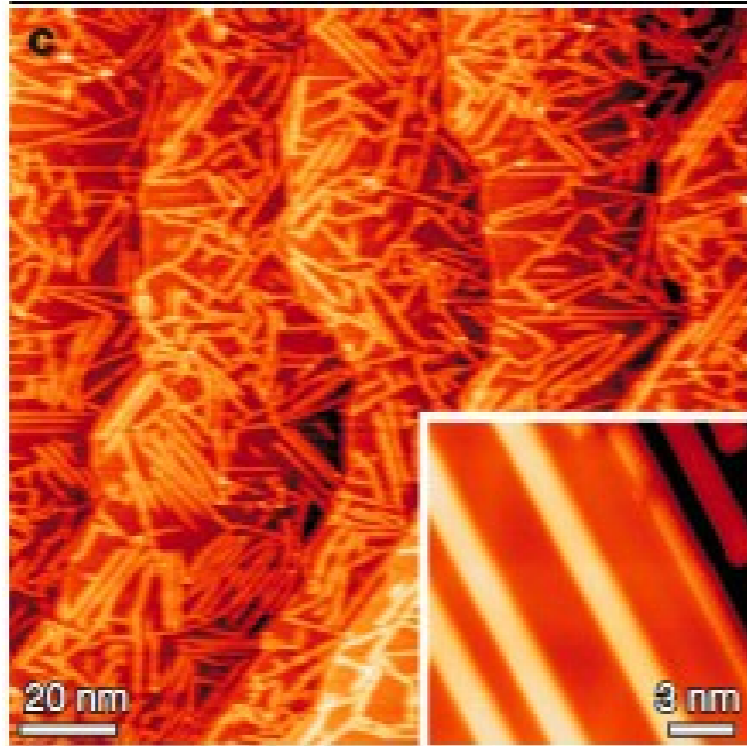
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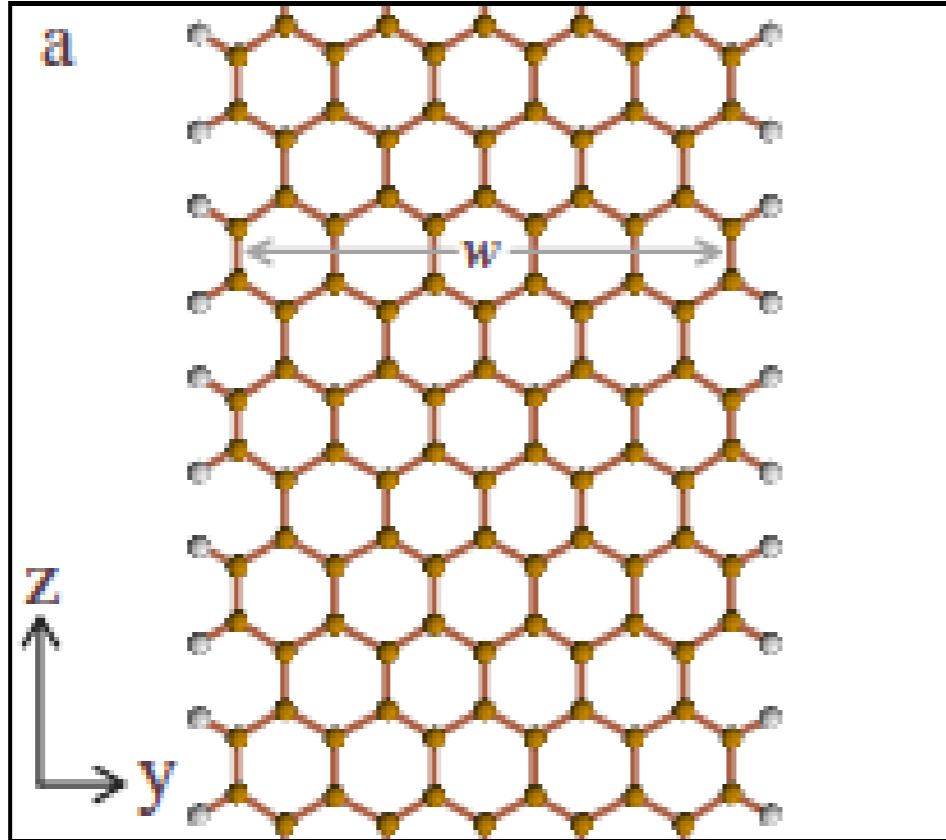
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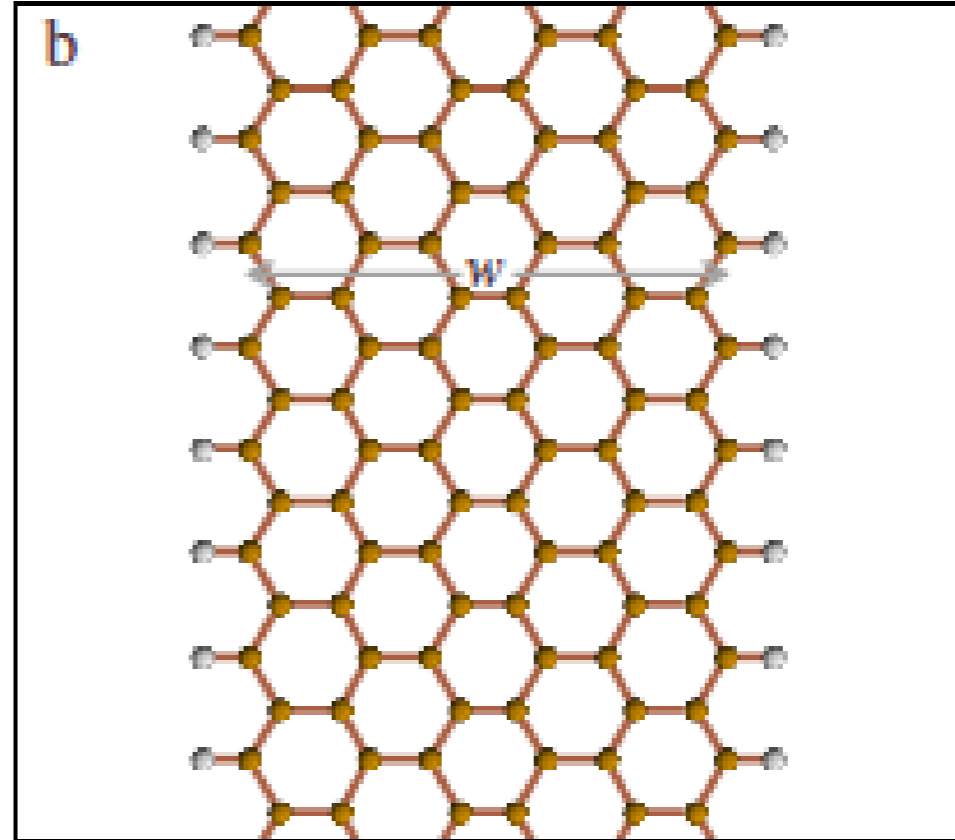


7-AGNR

Fasel, Müllen et al., Nature 2010



11-Armchair



6-Zig-zag

Louie et al., PRL 2007

Graphene International School, Cargese, October 11-22, 2010

Graphene from molecular precursors

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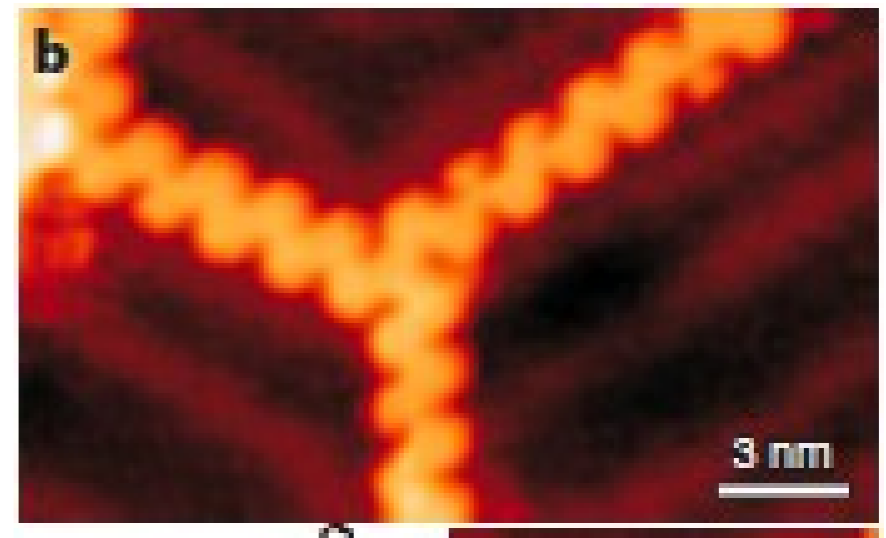
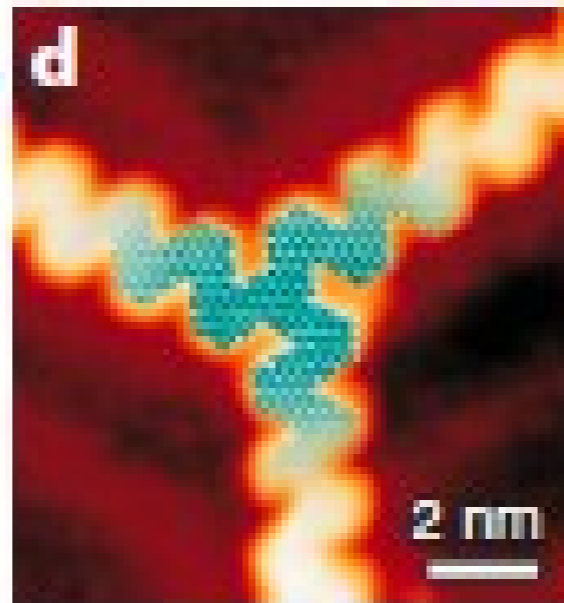
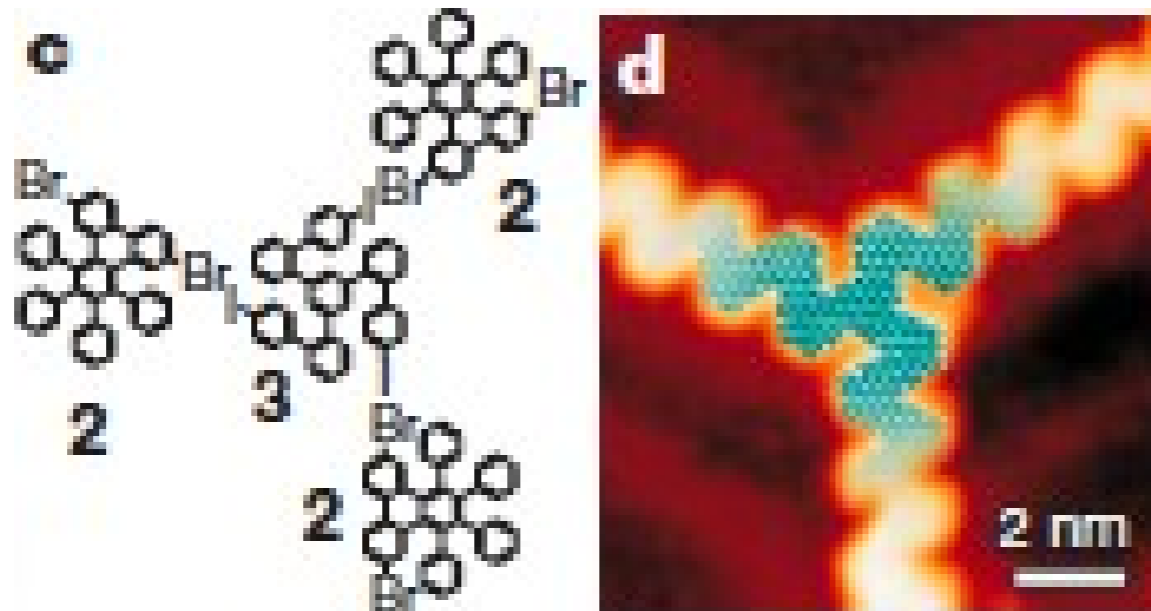
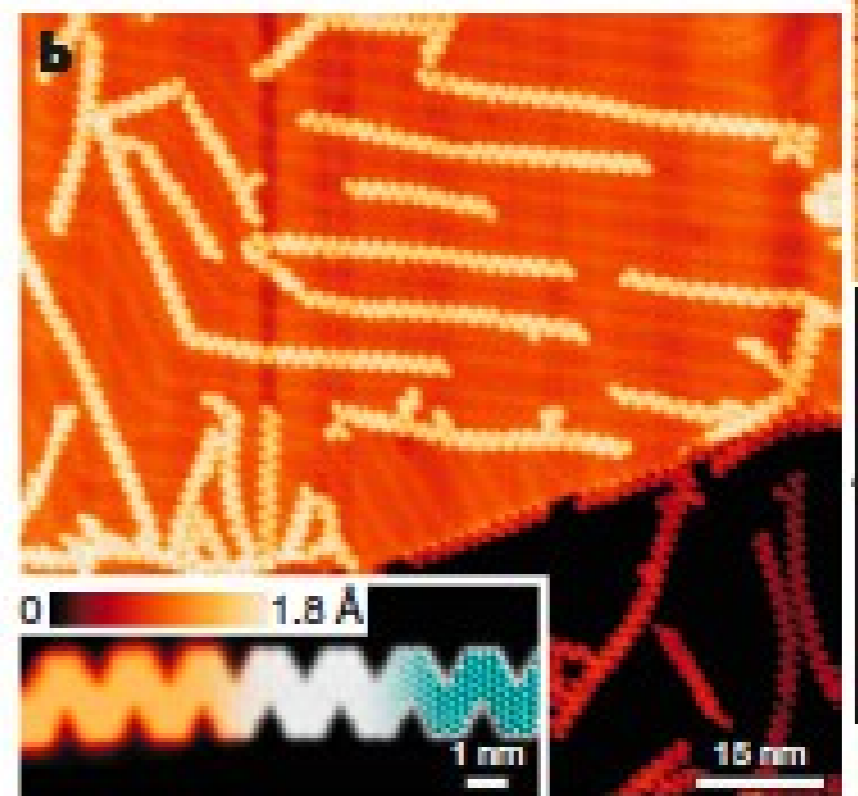
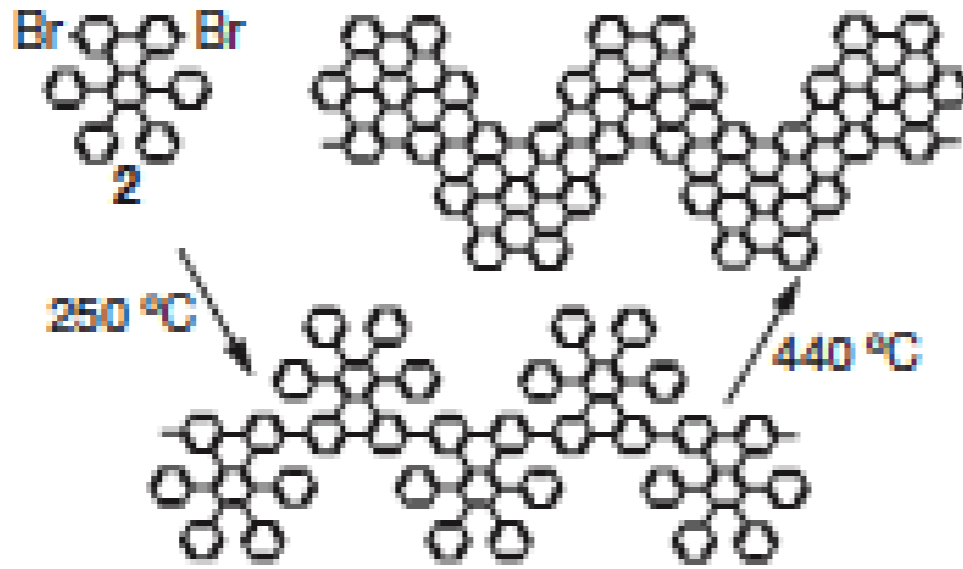
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